



RICE UNIVERSITY

School of Engineering

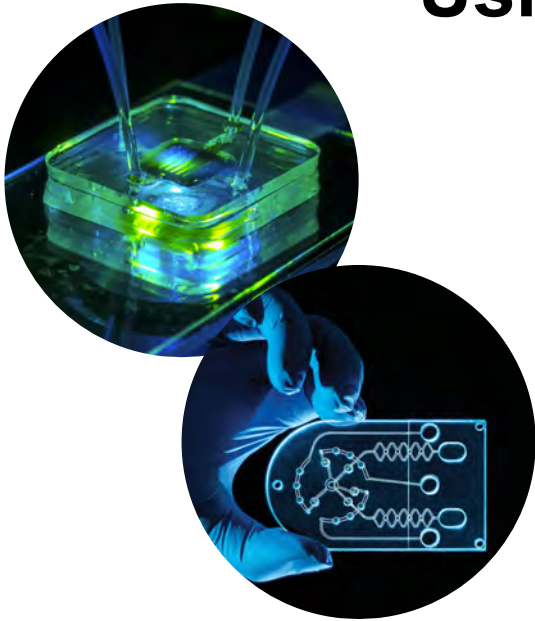
*Department of Chemical and Biomolecular Engineering*



# Physicochemical Characterization of Asphaltenes Using Microfluidic Porous Media

*Sibani Lisa Biswal*

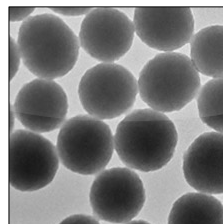
*William M. McCardell Professor and Chair  
Chemical and Biomolecular Engineering  
Rice University*



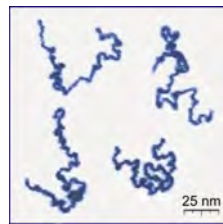
# Engineering Soft Matter

## Advanced Materials:

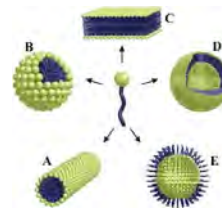
- Superparamagnetic Colloidal Crystals
- Flexible magnetic filaments
- Micromechanical sensors



Colloids



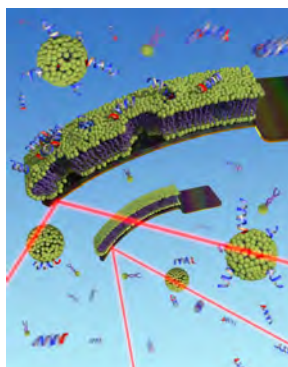
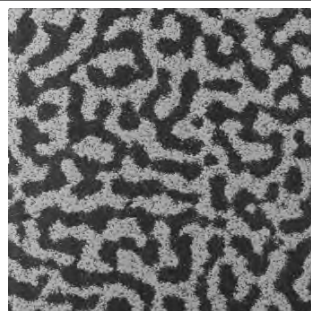
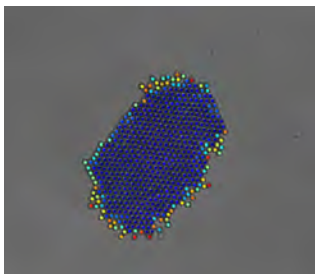
Polymers



Surfactants

## Energy Applications:

- Bubbles and Foams for CO<sub>2</sub> sequestration
- Emulsions for remediation
- Silicon Anodes for Lithium-Ion Batteries
- Electrochemical separation of lithium from geothermal brines

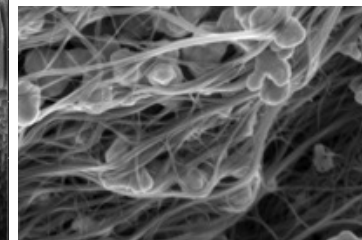
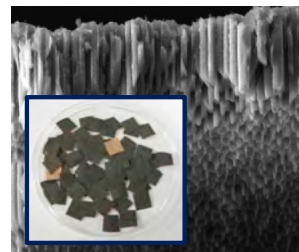
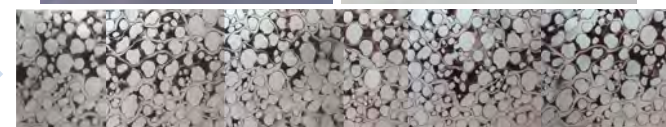
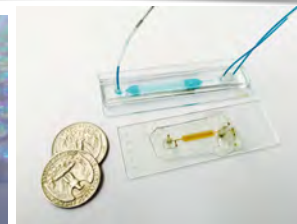
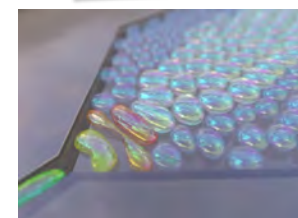


Directed Paramagnetic Colloidal Assemblies

Multiphase Flows in Porous Media

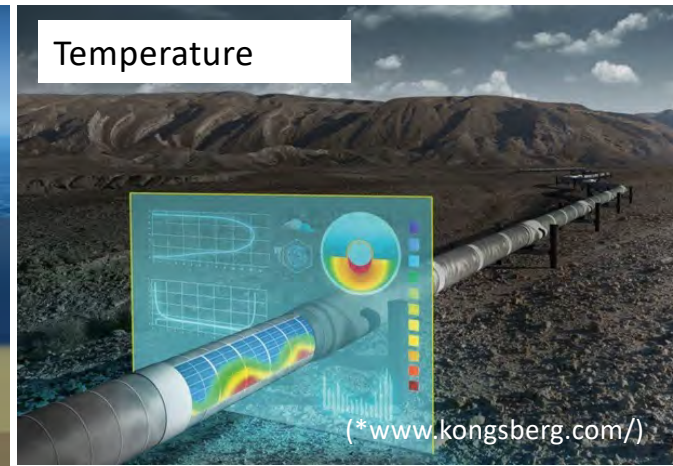
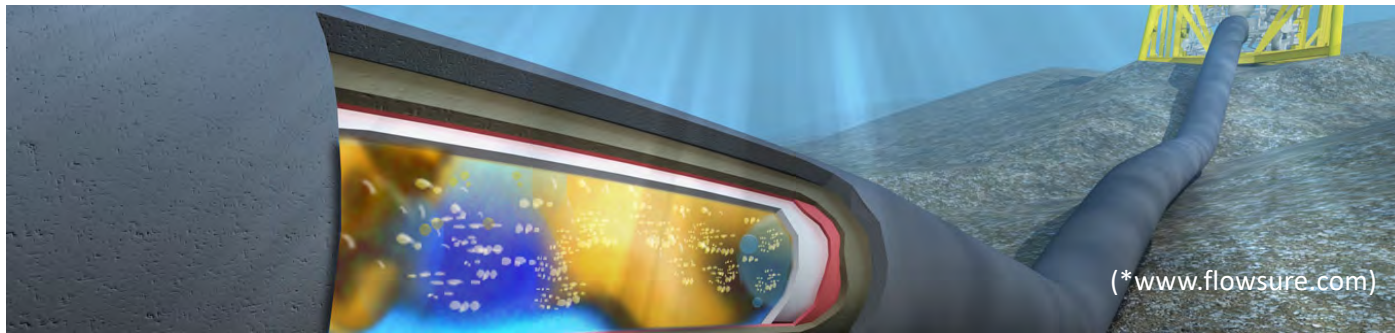
Lipid and Protein-Based Biosensors

Porous Silicon Composites for Batteries

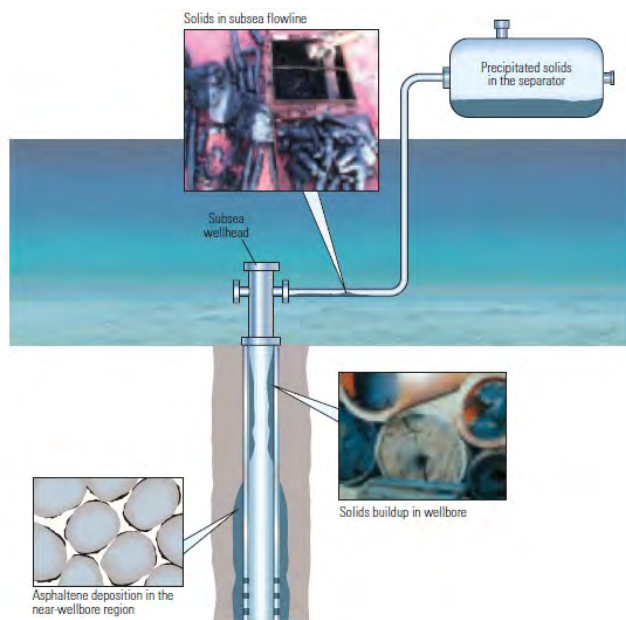


# Multiphase Fluid Flows

Complexities of multiphase flows is fundamental to the success of oil and gas operations, impacting everything from production and transport to safety and environmental protection.



# Coating Problem: Asphaltenes

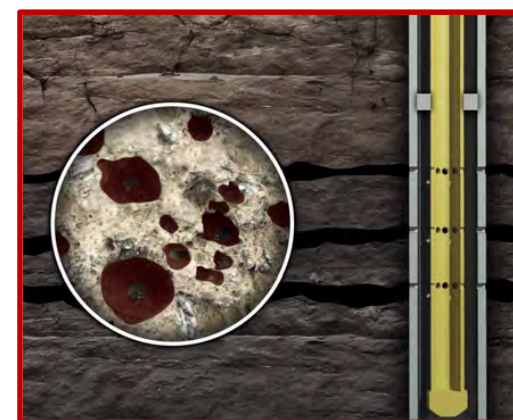


Plugging in the wellbore



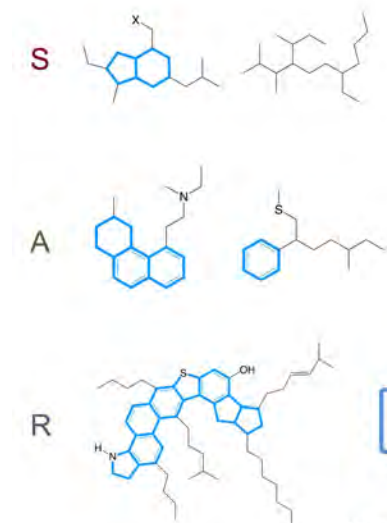
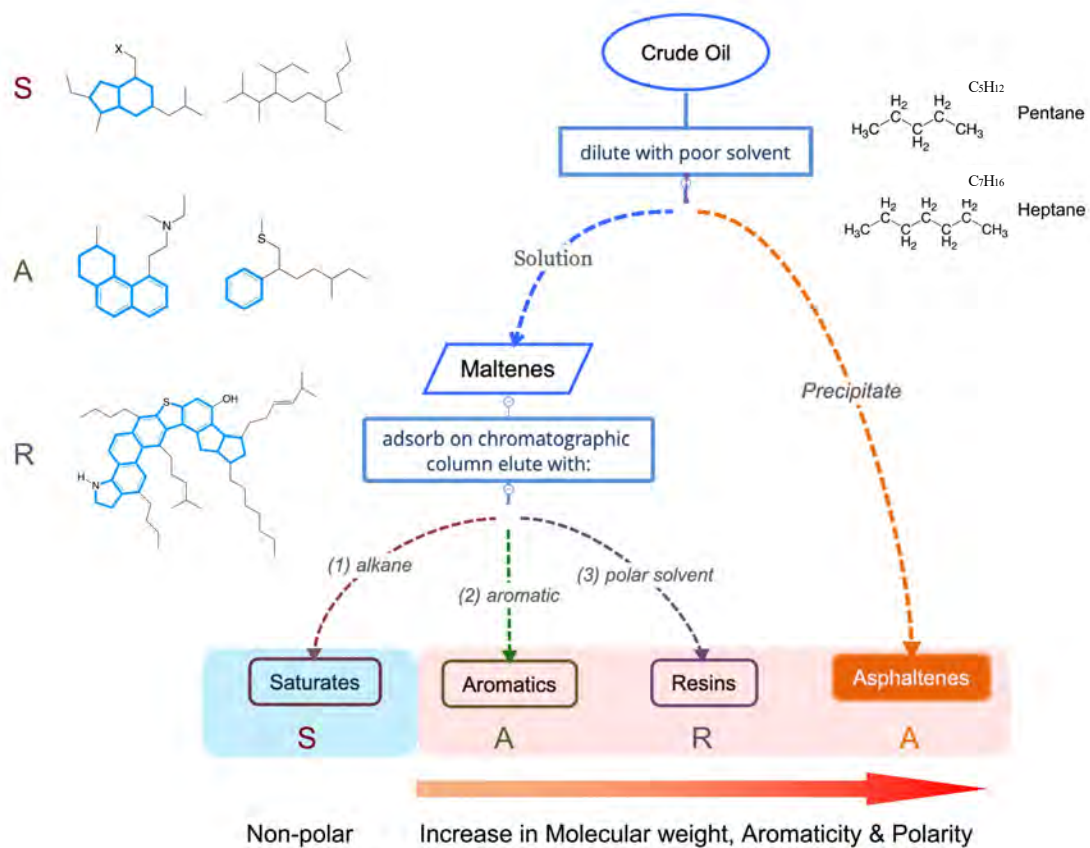
High cost of remediation:  
 ~ \$500 k/well onshore  
 ~ \$3 MM/well offshore  
 ~ 1.2 MM per day  
 (lost production)

Plugging in the porous media



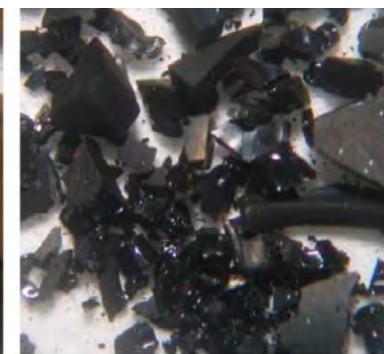
(\*Oilfield review, 2007)

# What are Asphaltenes?

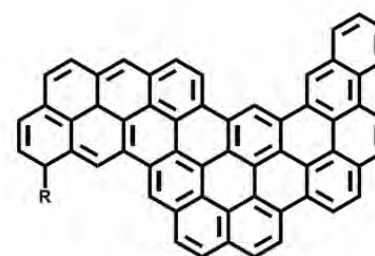


C5 Asphaltene

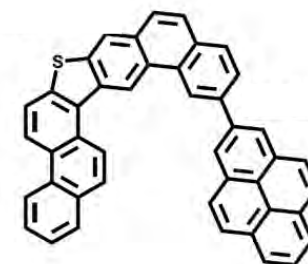
(\*Oilfield review, Summer 2007)



C7 Asphaltene



Island Motif

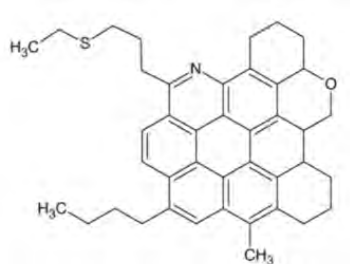
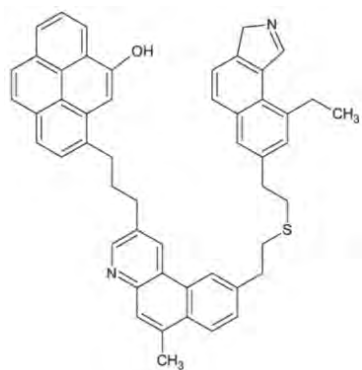


Archipelago Motif

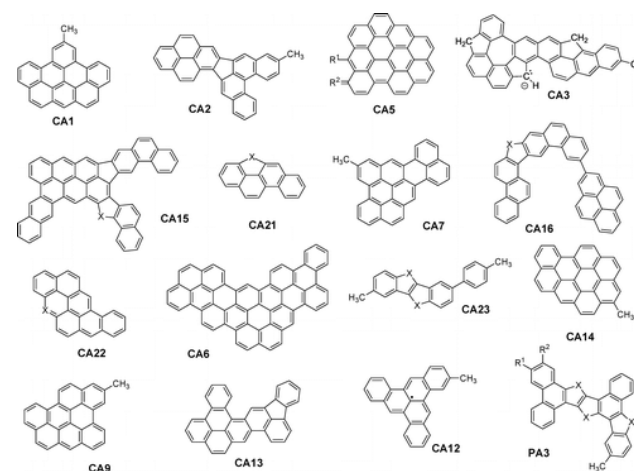
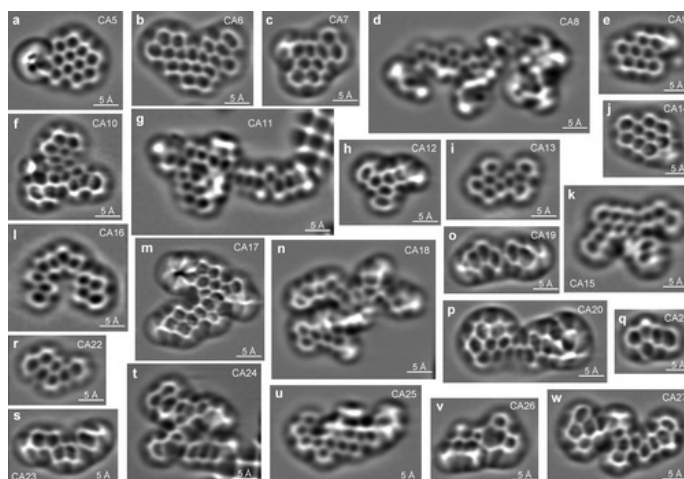
(\*M. L. Chacon *et al.*, Florida State University)

(\*Jill Buckley's Lab, New Mexico Tech)

# What are Asphaltenes?

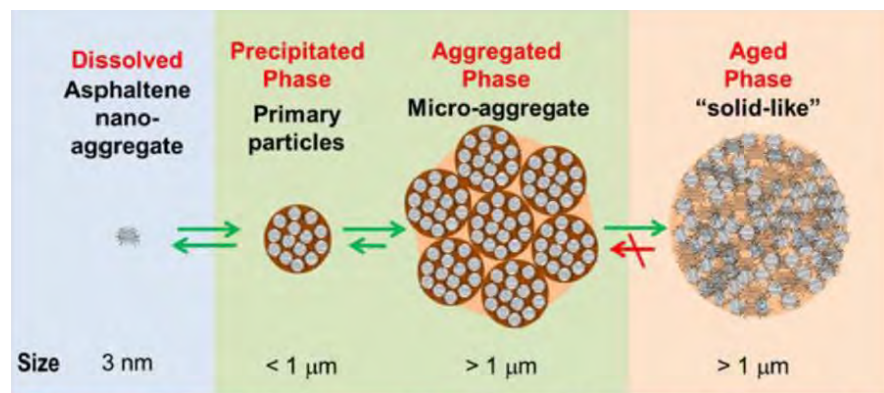
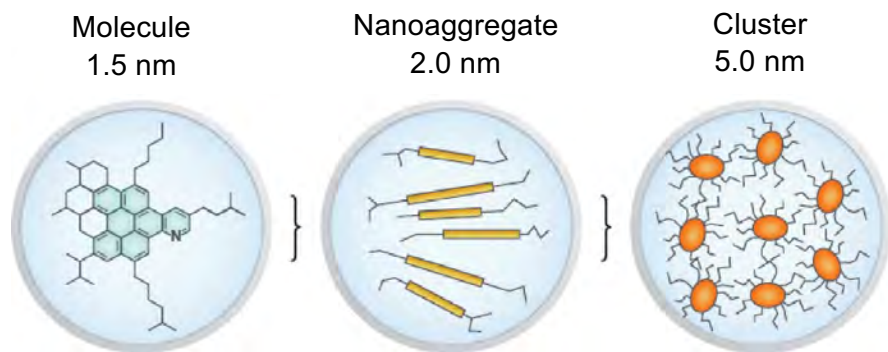


- Asphaltene chemical structure generally contains:
  - Fused ring structures
  - Naphthenic rings
  - Aliphatic chains
    - Hydrogen and carbon
    - Heteroatoms (nitrogen, sulfur, oxygen)
    - Trace metals (vanadium, nickel)

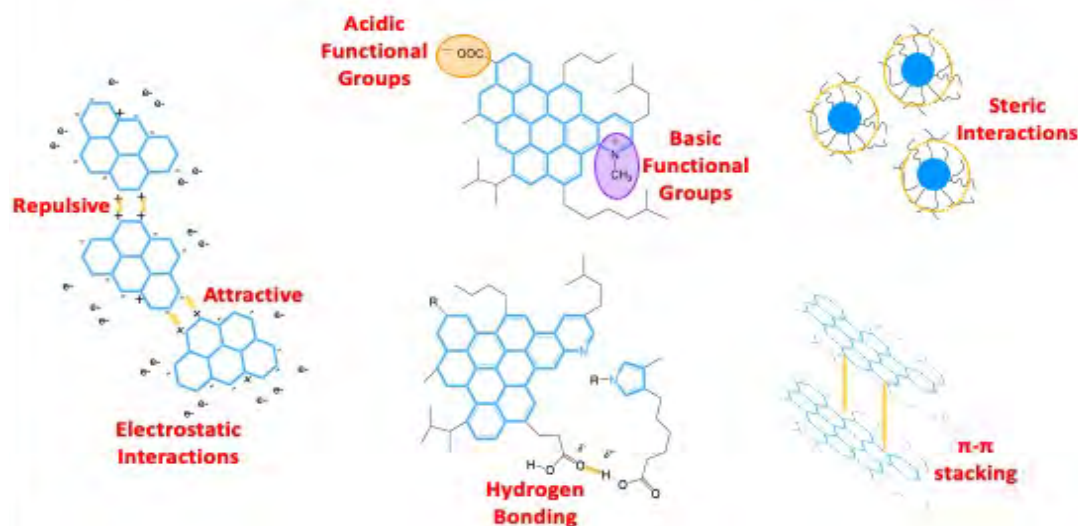


# Asphaltene Physicochemical Properties

## Colloidal behavior of asphaltenes

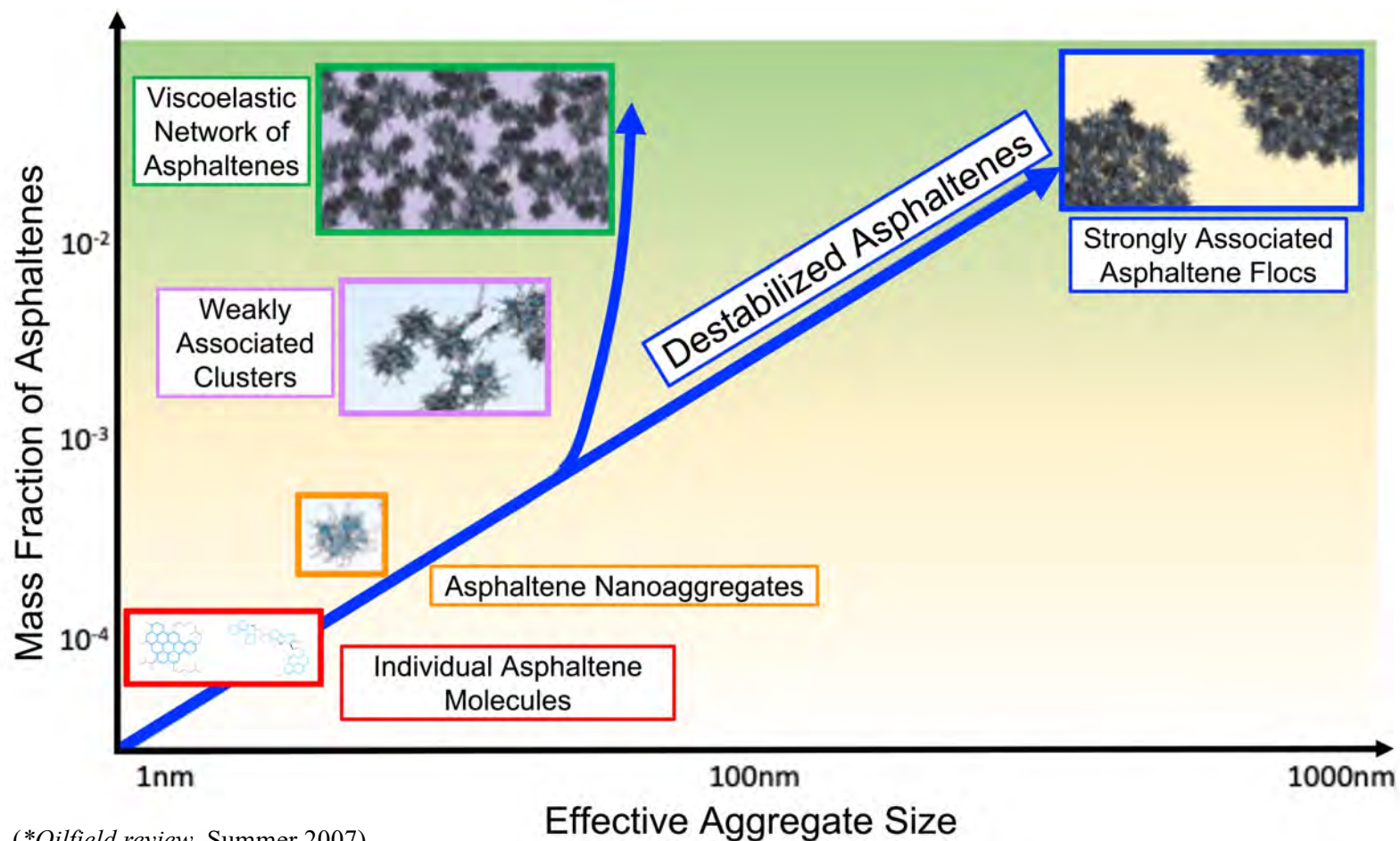


## Key governing intermolecular forces



(\*Mullins *et al.*, **Energy Fuels** 2012, 26, 3986–4003)

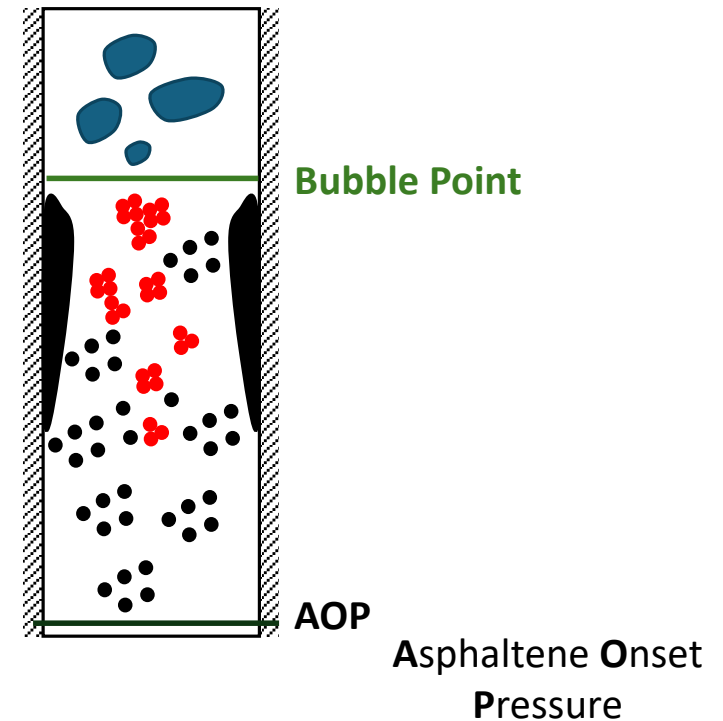
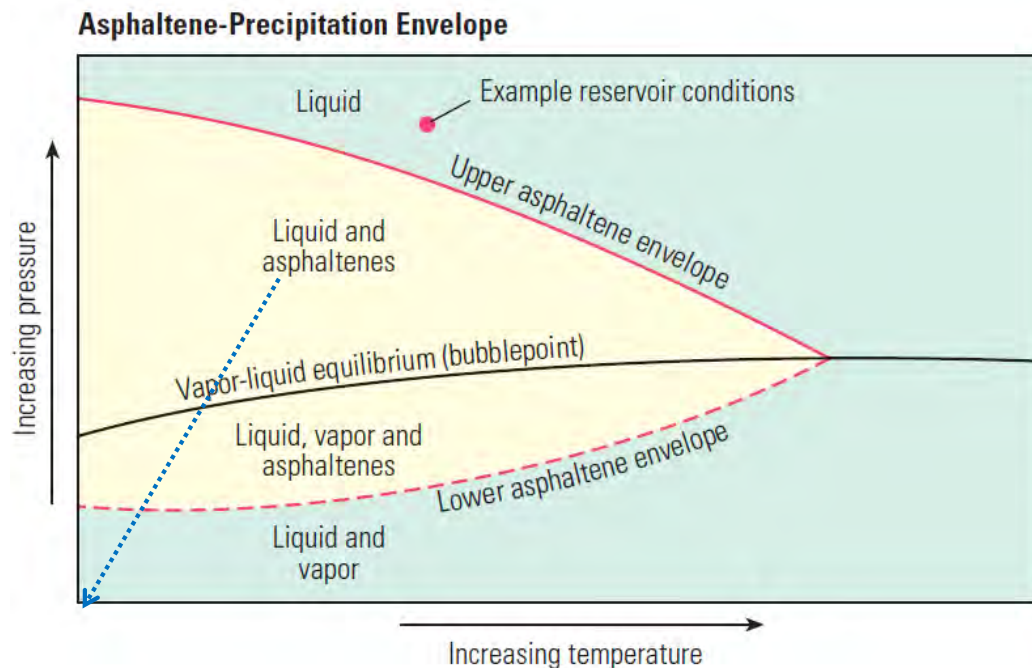
# Asphaltene Physicochemical Properties



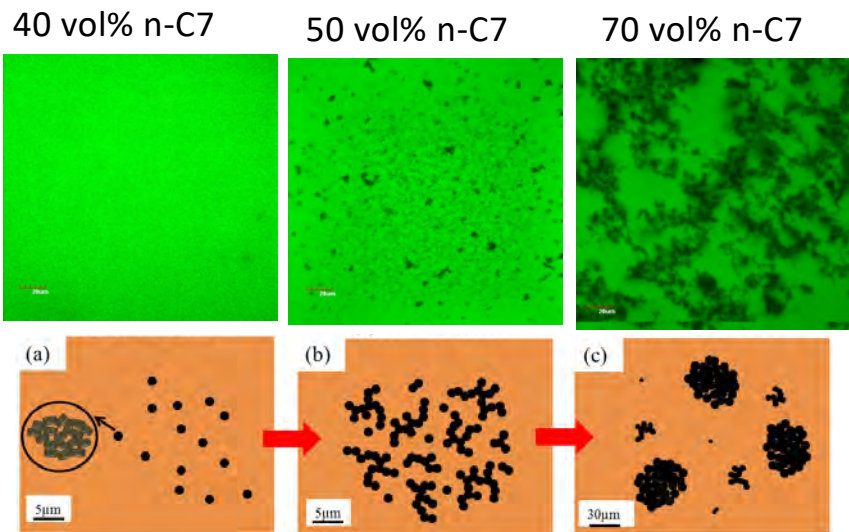
(\*Oilfield review, Summer 2007)

# Asphaltene Destabilization & Precipitation

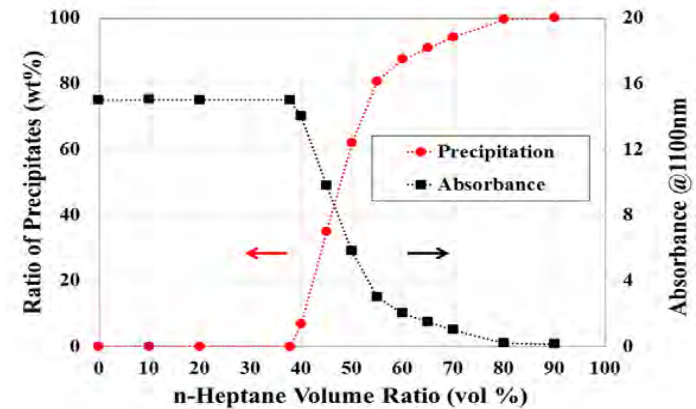
- Precipitate due to pressure, temperature, and composition change
- Insoluble: *n*-alkane (*n*-heptane and *n*-pentane)
- Soluble: aromatics (toluene and xylene)



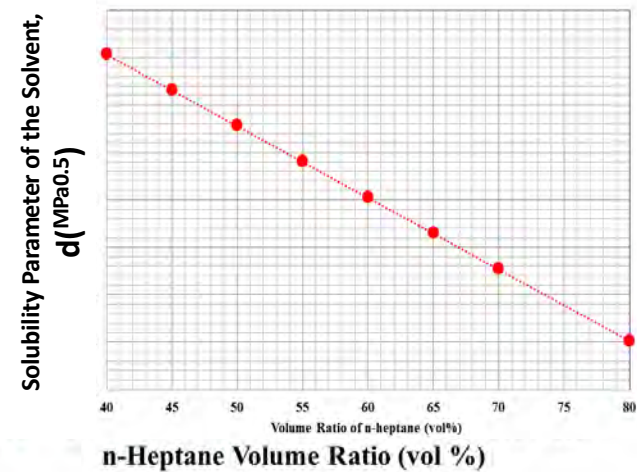
# Asphaltene Aggregation Model



4.6wt% asphaltene in a model oil

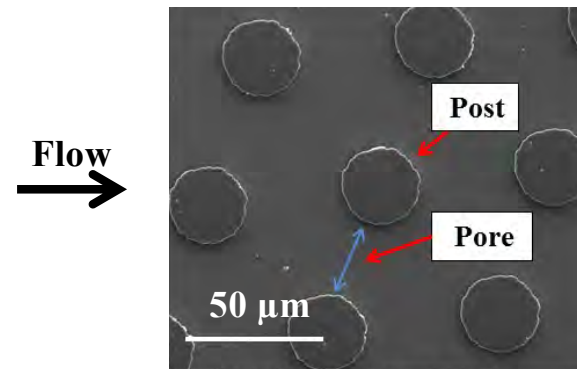
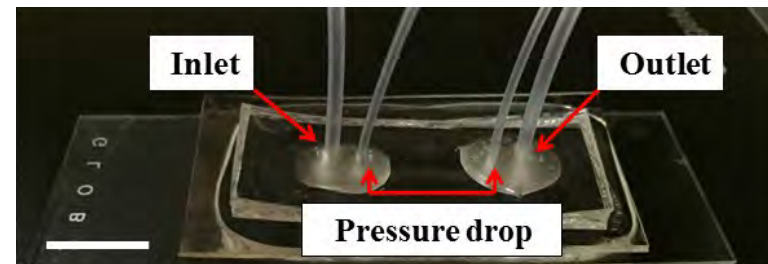


Soluble Insoluble



(\*Lin, Biswal, Langmuir, 2016)

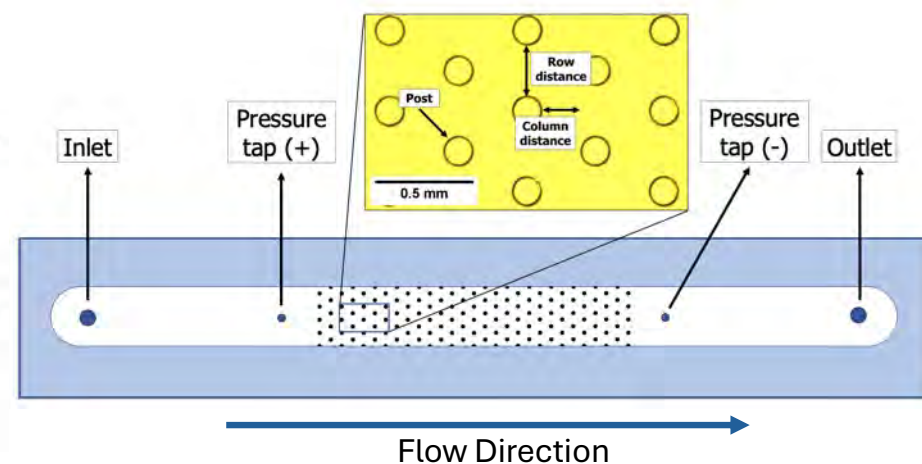
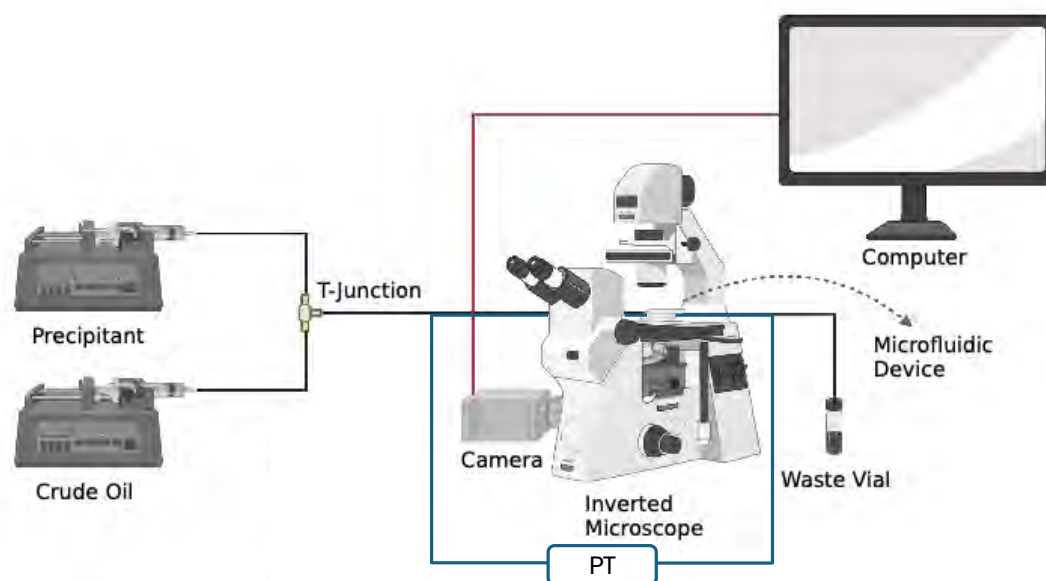
# Microfluidic Model of Porous Media



(\*Yu-Jiun Lin *et al.*, *Langmuir*, 2016)

- Behavior of asphaltenes are detected both macroscopically (pressure drop) and in the pore scale (visualization).
- Knowledge of asphaltenes at the *micro scale* helps to solve related flow assurance issues at *field scale*

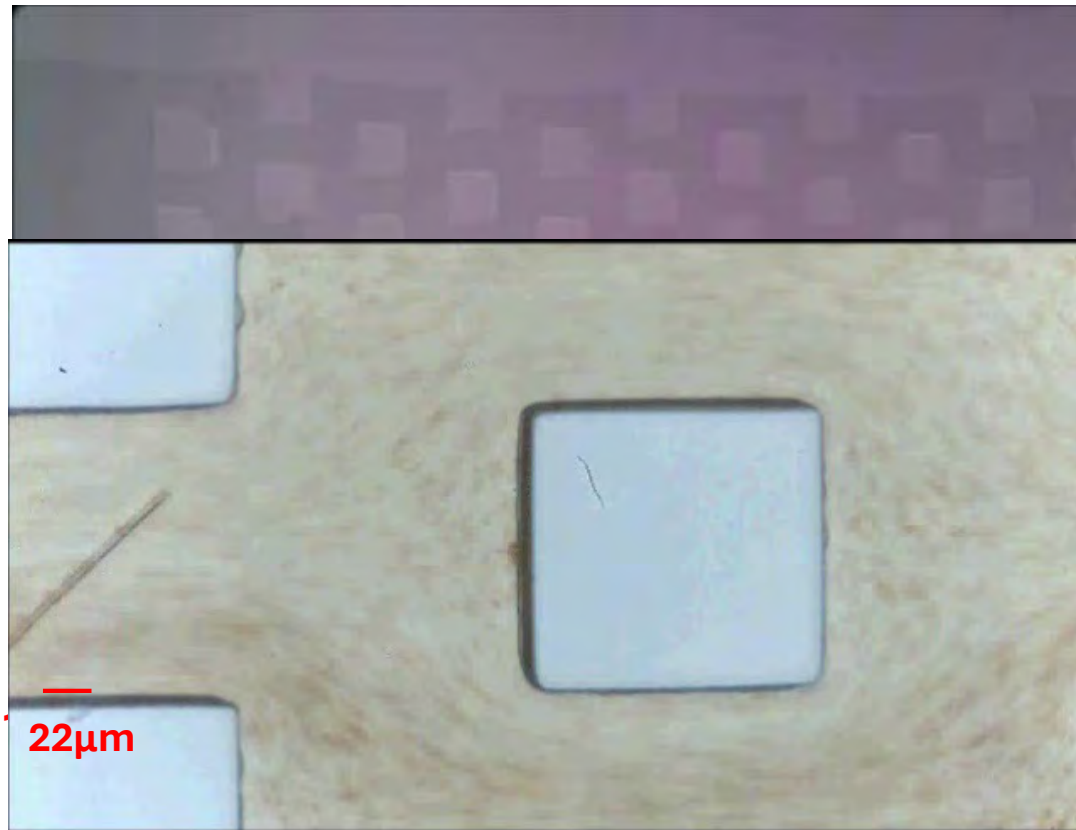
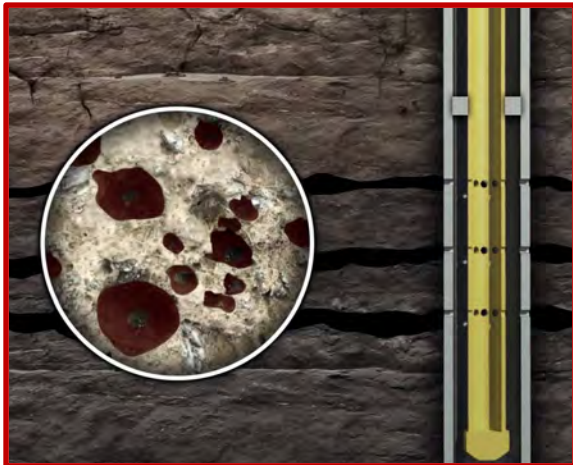
# Microfluidic Experimental Setup



|                          |                   |
|--------------------------|-------------------|
| Porous media's length    | 1 mm              |
| Porous media's width     | 0.2 mm            |
| Porous media's thickness | 40 $\mu\text{m}$  |
| Post's diameter          | 120 $\mu\text{m}$ |
| Row distance             | 280 $\mu\text{m}$ |
| Column distance          | 220 $\mu\text{m}$ |
| Porosity                 | 91.6%             |
| Permeability             | 40 Darcy          |

# Asphaltene Deposition in Porous Media

1. n-C7 70%, Oil 30%
2. Ambient Conditions

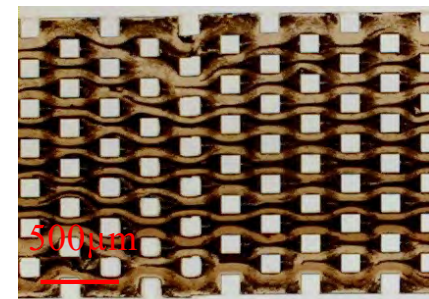
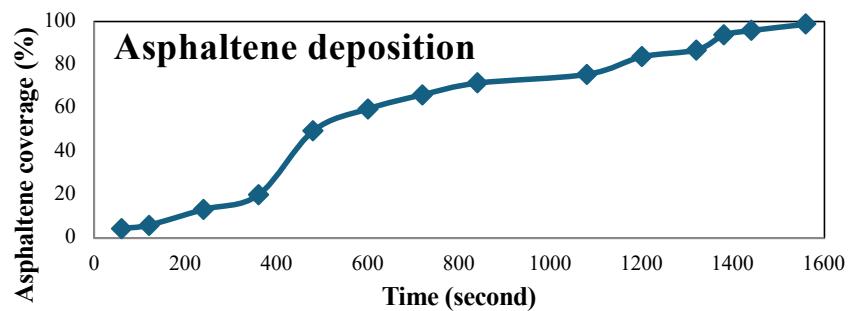


Flow

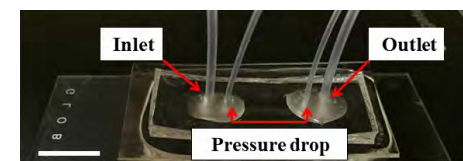
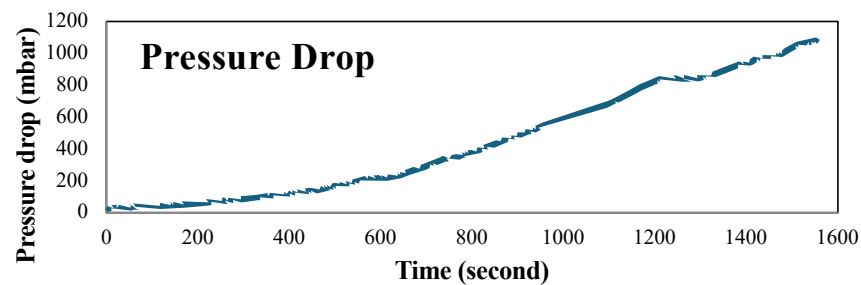


# Asphaltene Data

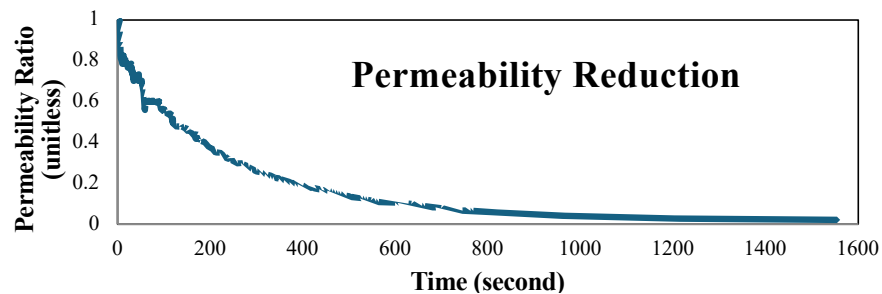
Imaging  
Process



Pressure  
transducer



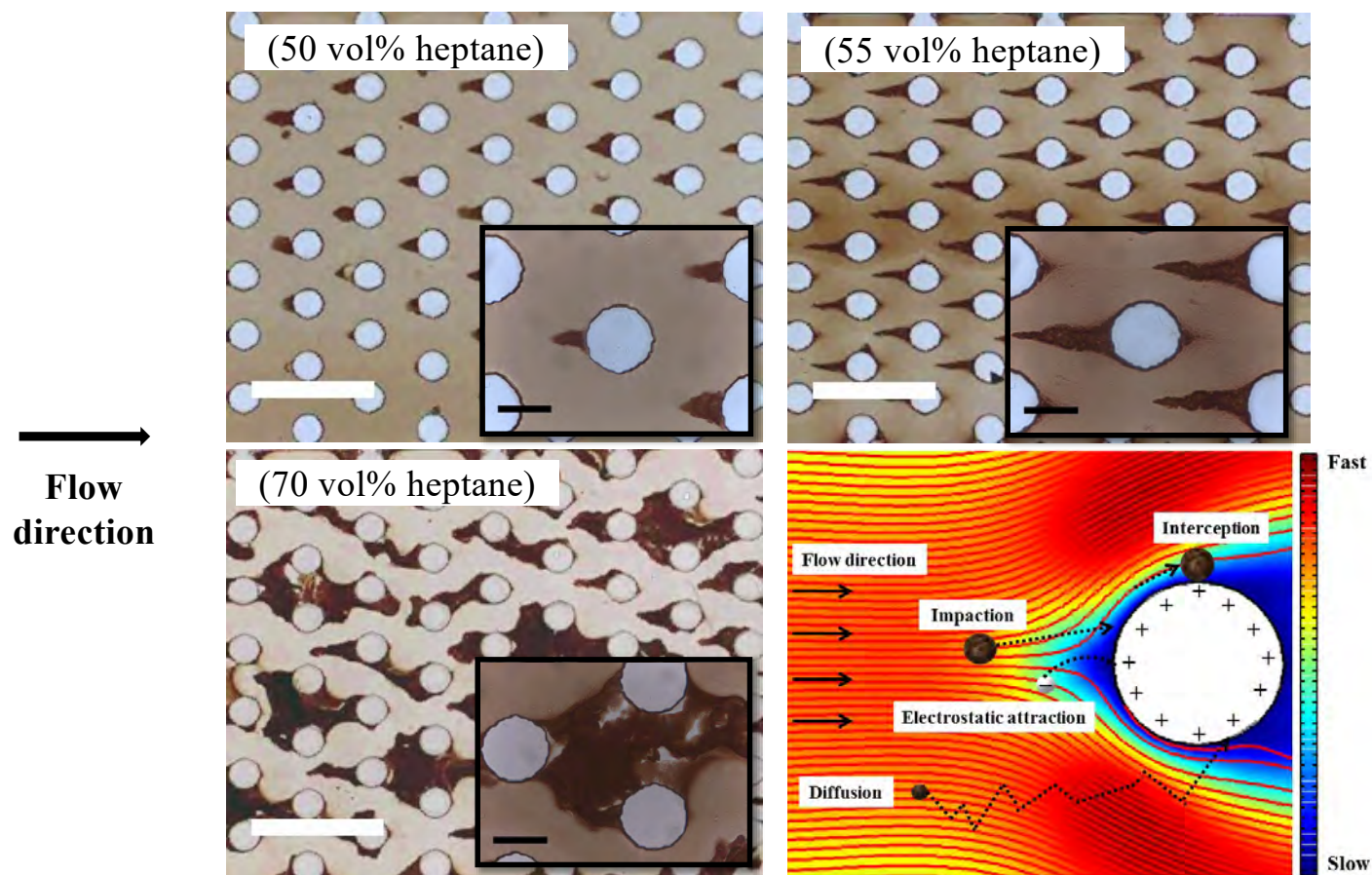
Calculation



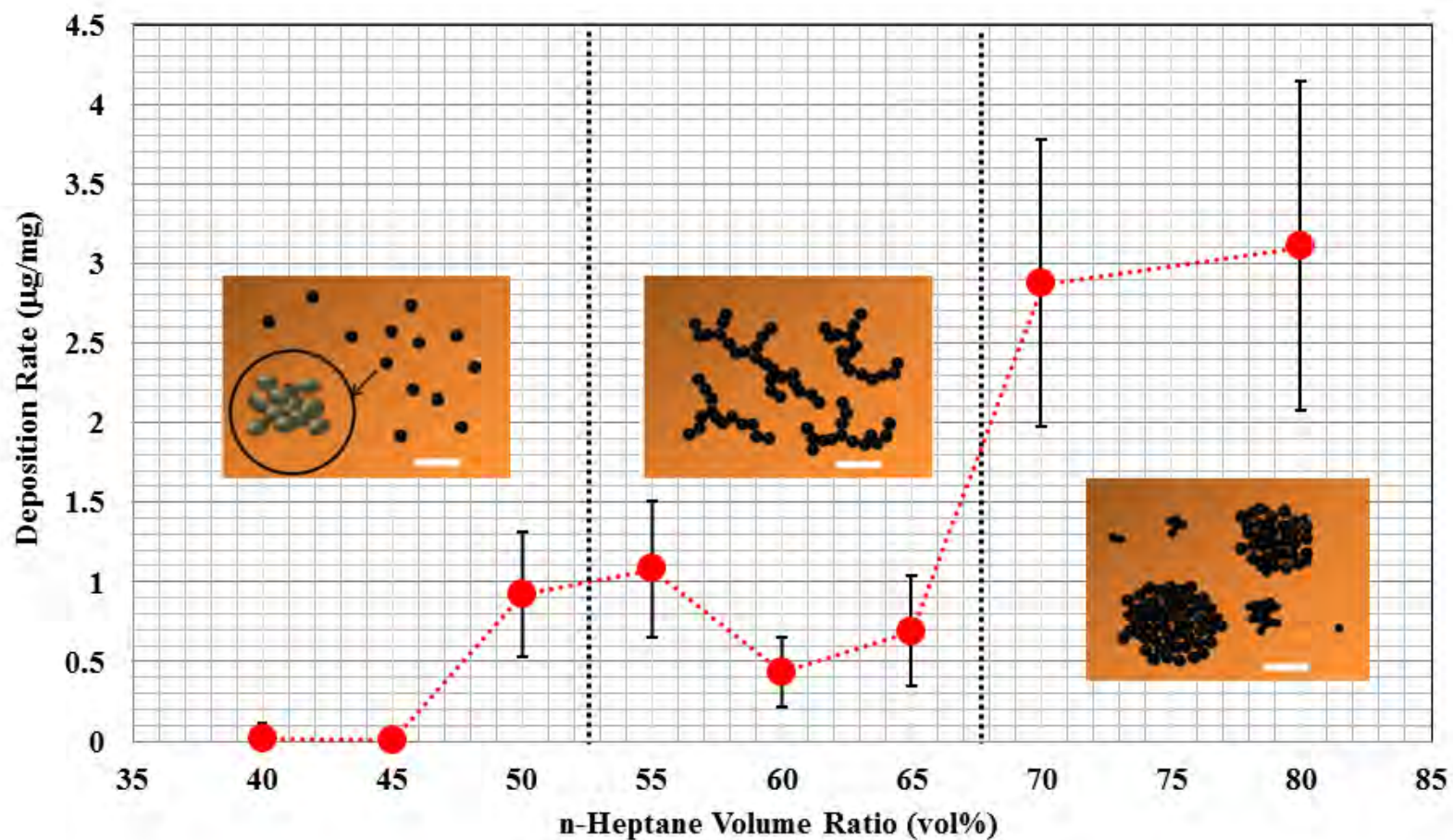
Darcy's Law:

$$Q = \frac{-kA \Delta P}{\mu L}$$

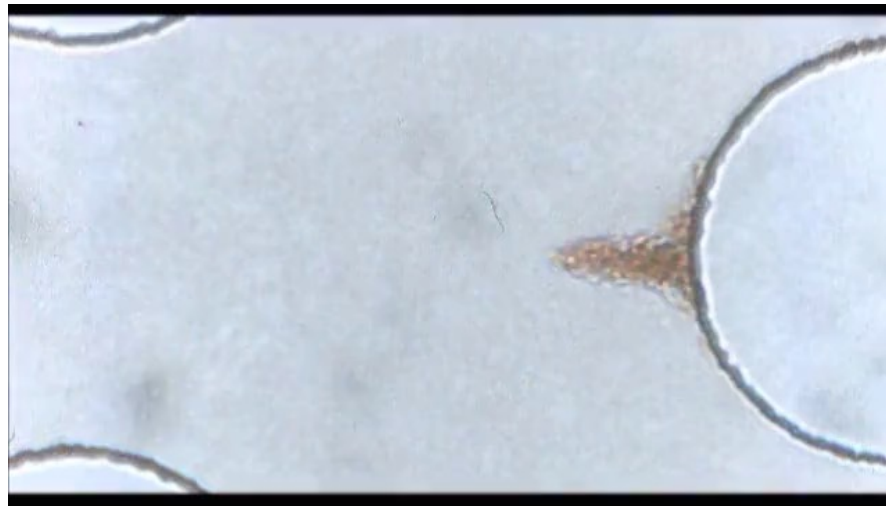
# Asphaltene Deposition in Porous Media



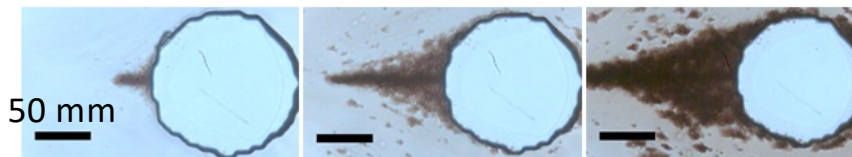
# Asphaltene Deposition Rate



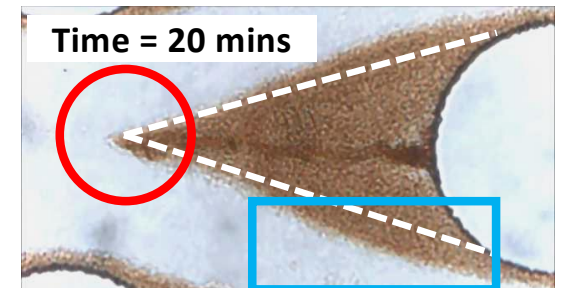
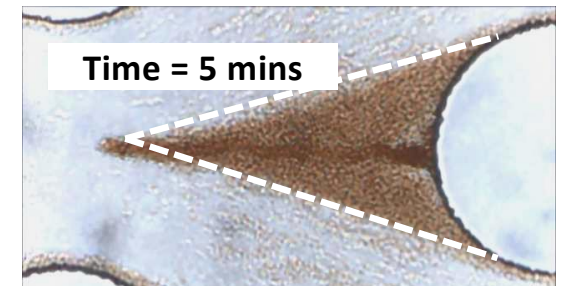
# Asphaltene Deposition in Porous Media



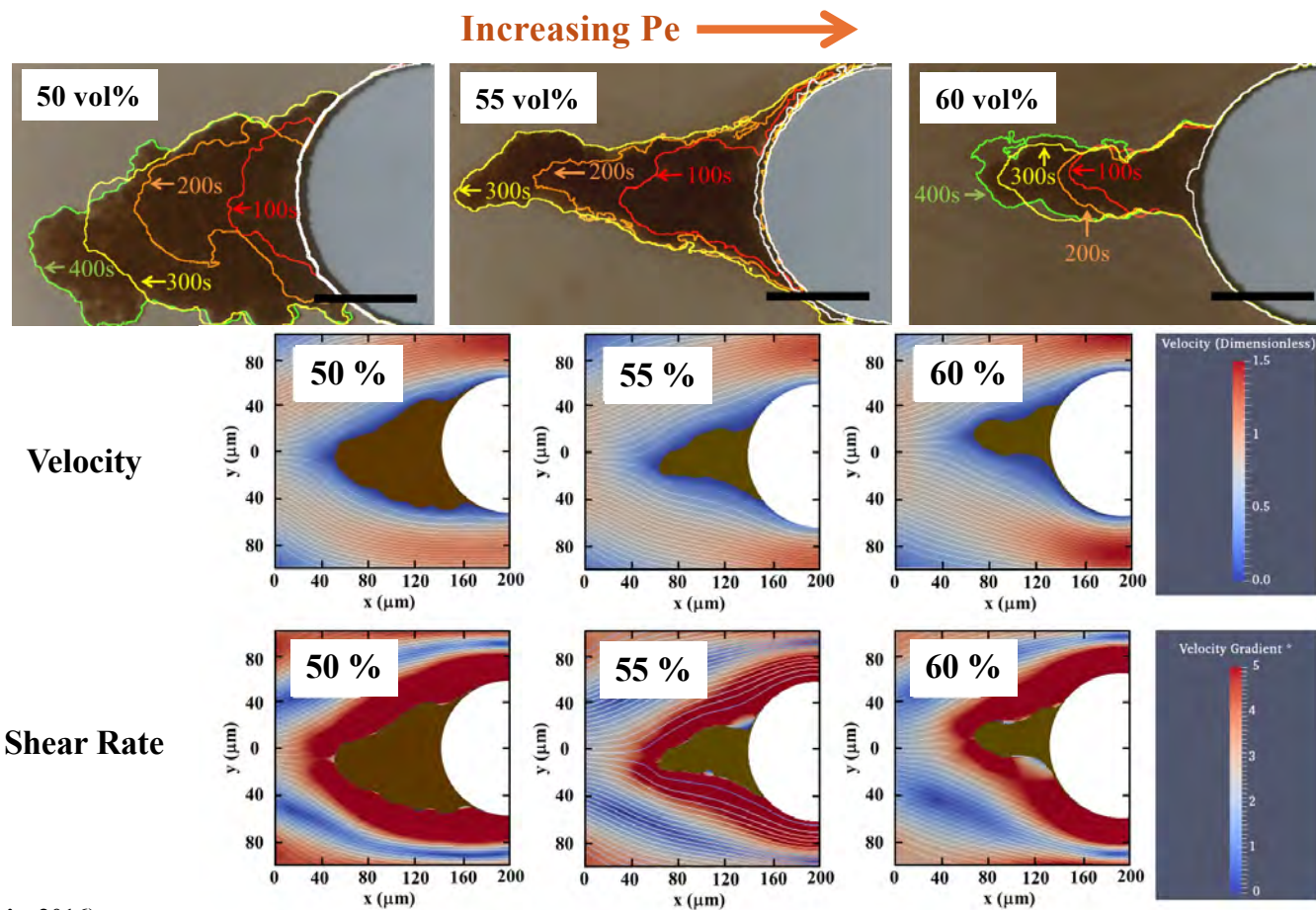
Flow direction



- The front of the deposition reaches a max growth at a critical shear rate
- However, the width of the deposition grows



# Pore Scale Velocity and Shear Rate



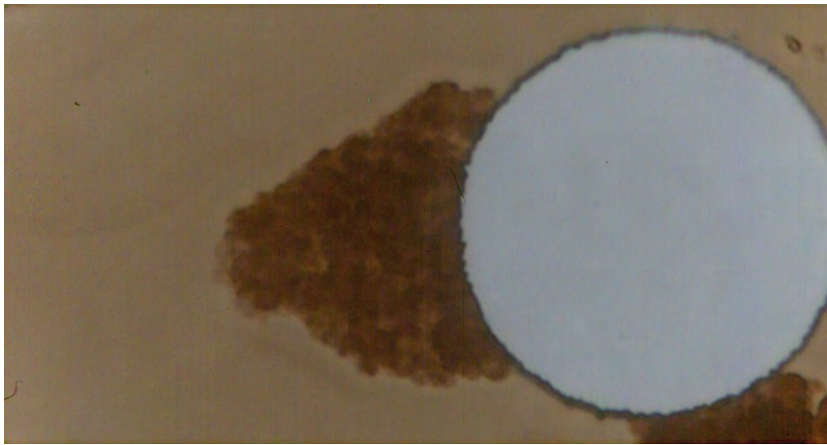
(\*Yu-Jiun Lin *et al.*, *Langmuir*, 2016)

# Asphaltene Aggregation Structure

$Pe = 10^4 - 10^5$

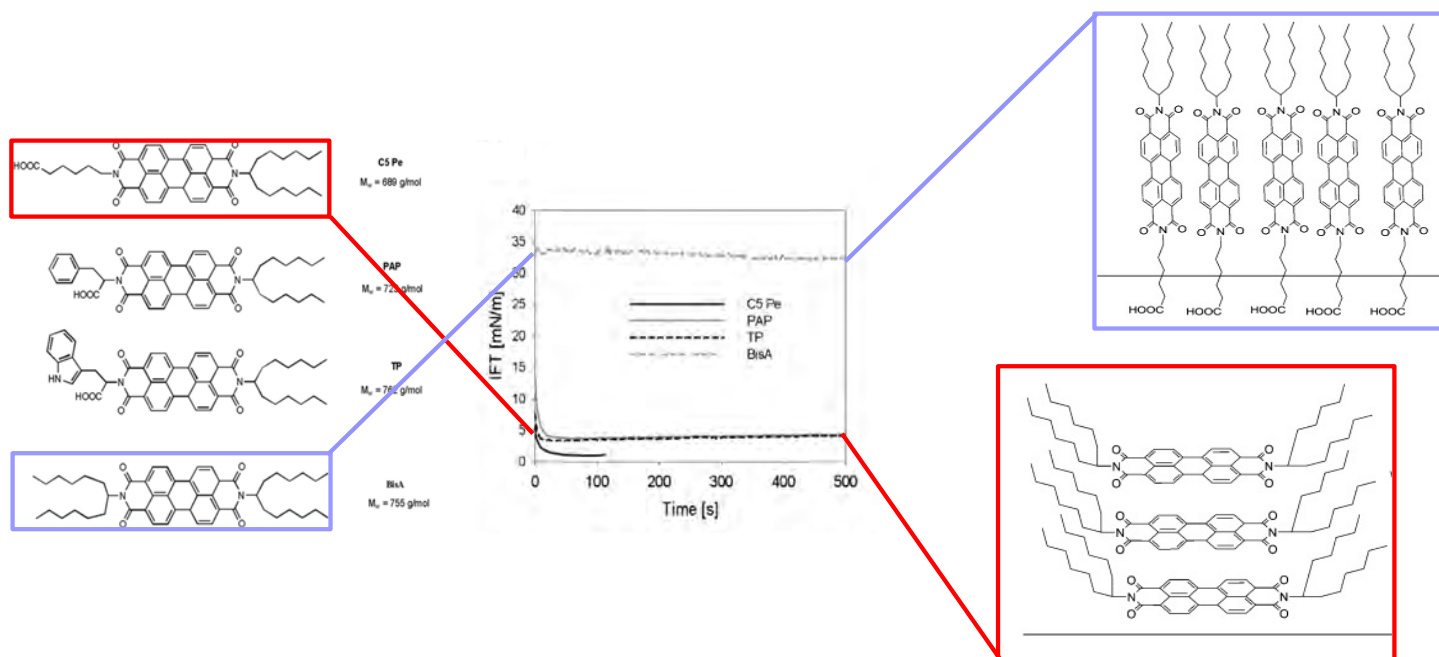
$$Pe = \frac{\textit{Convection}}{\textit{Diffusion}}$$

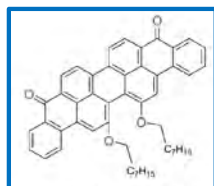
$Pe = 10^5 - 10^6$



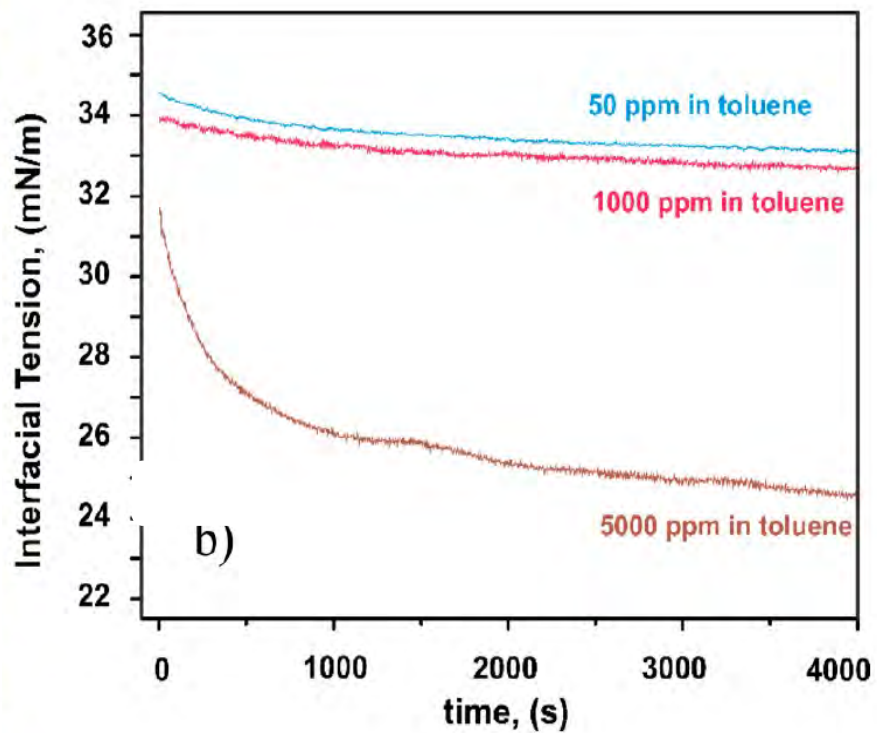
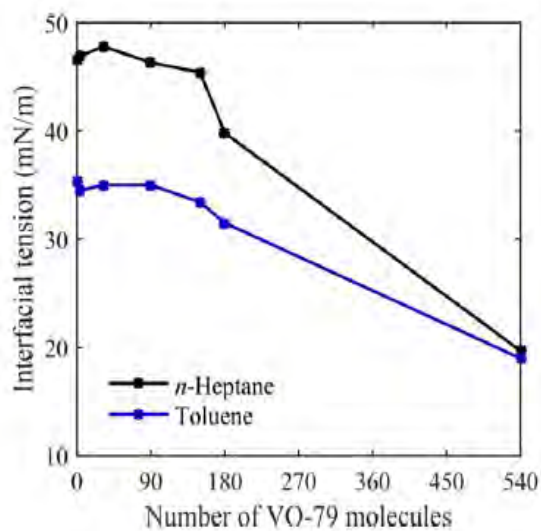
Deposition profile a function of Peclet number, which is a function of solubility

# Model Asphaltene Molecules





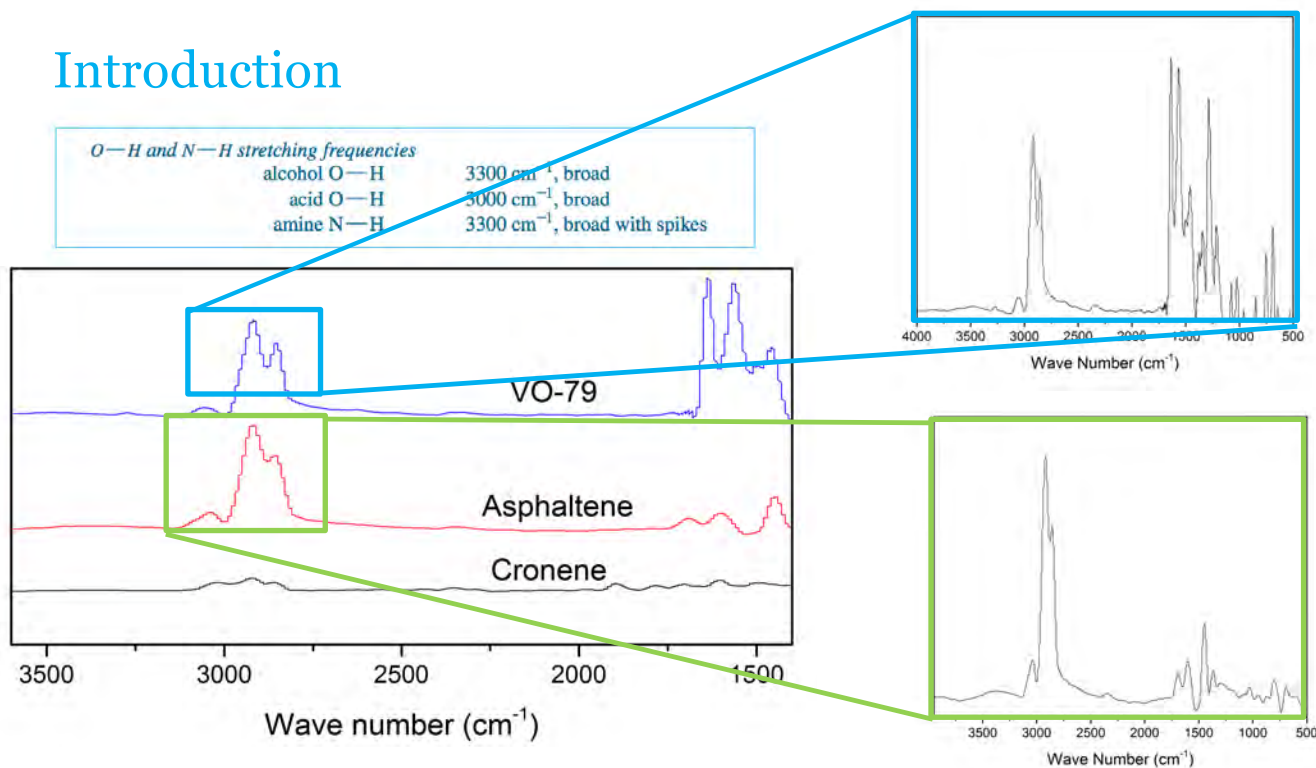
VO-79



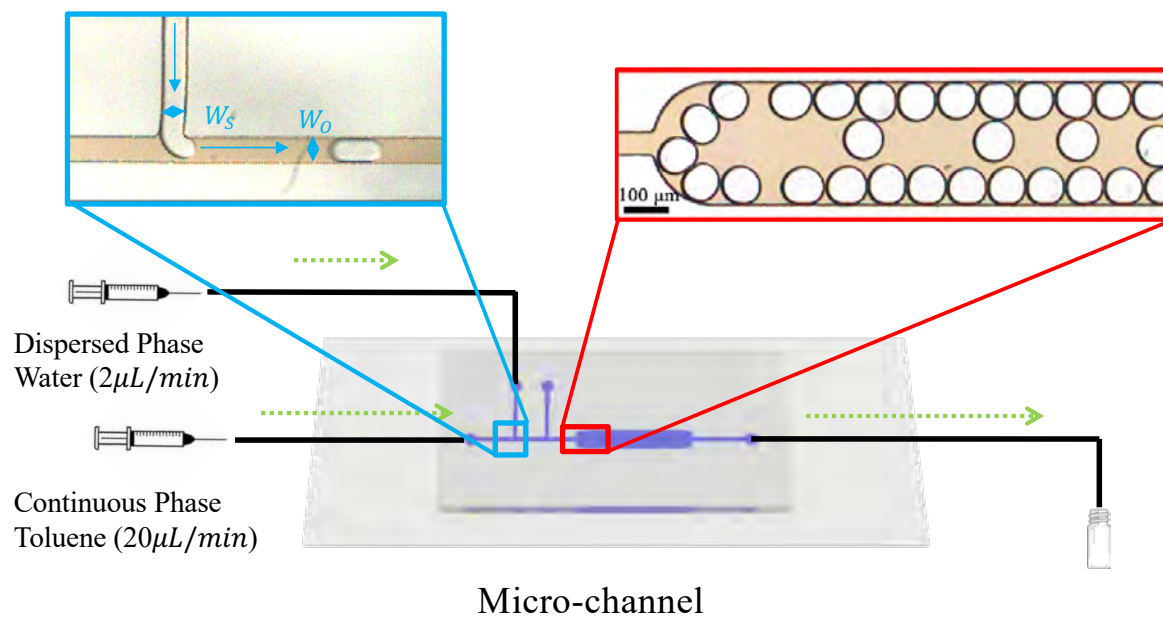
## Introduction

*O—H and N—H stretching frequencies*

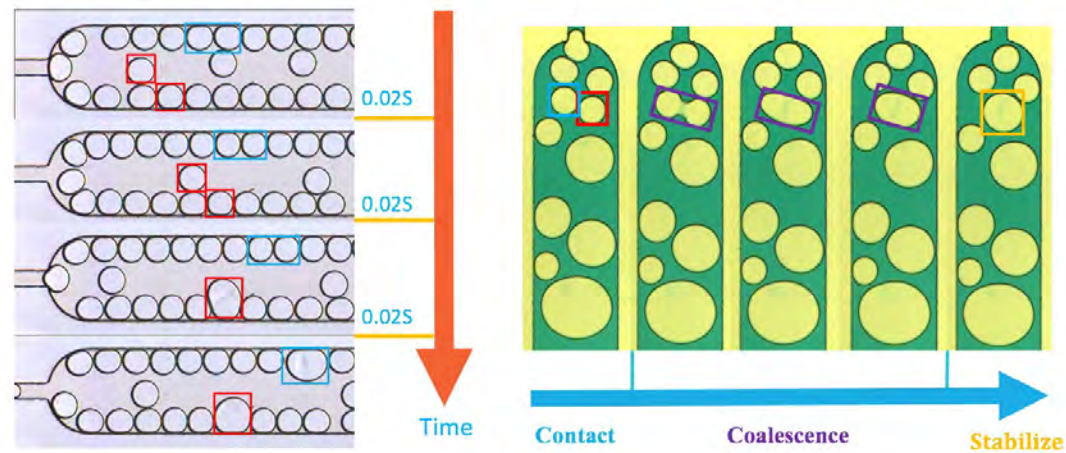
|             |   |
|-------------|---|
| alcohol O—H | 3300 $\text{cm}^{-1}$ , broad             |
| acid O—H    | 3000 $\text{cm}^{-1}$ , broad             |
| amine N—H   | 3300 $\text{cm}^{-1}$ , broad with spikes |



## Probing Emulsion Stability Using Microfluidic Devices



## Coalescence Phenomenon Inside Micromodel



$$R_c = \frac{n_c}{A_c t_{exp} n_T}$$

$$n_c = \sum_i^j n_{c,ij}$$

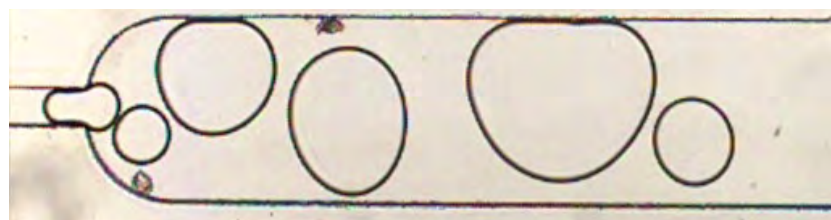
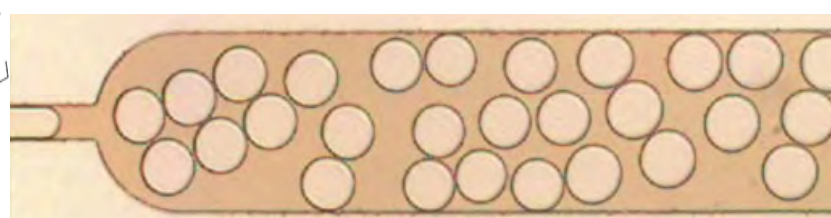
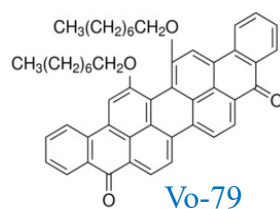
$R_c$  - Coalescence rate ( $m^{-2} \cdot s^{-1}$ )

$A_c$  - Monitored area ( $m^2$ )

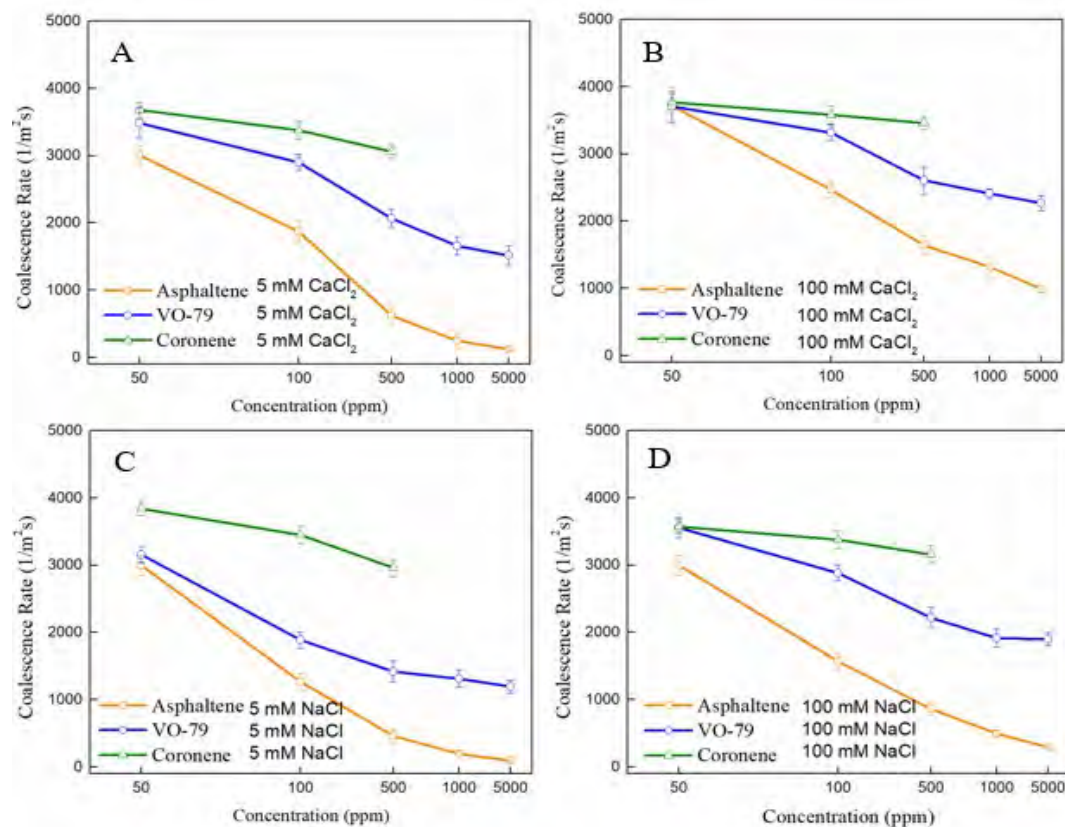
$n_c$  - number of coalescence events

$n_T$  - total number of initial emulsions

## Coalescence Behavior Inside Collision Chamber




## Effect of salt relate to emulsion stability




- Coronene stabilized emulsions are not affected by salt concentration
- Asphaltene stabilized emulsions are more sensitive to salt concentration
- Asphaltene stabilized emulsions are more stable than that of VO-79 in both high and low salt concentration

# Emulsion Surface Characterization


## Zeta Potential Measurement




Emulsion Stability



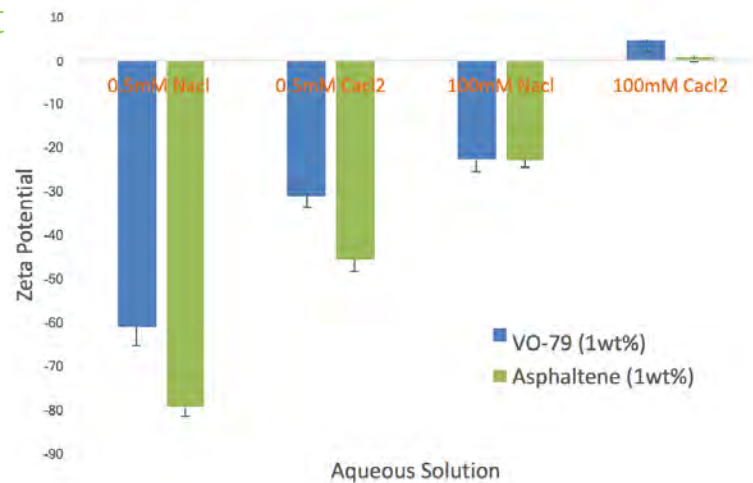
C1=CC=C2C(=C1)C=CC=CC2  
**Coronene**



CC1=CC=C2C(=C1)C(=O)C=C(C2)C3=CC=CC=C3C4=CC=CC=C4C5=CC=CC=C5C6=CC=CC=C6C7=CC=CC=C7C8=CC=CC=C8C9=CC=CC=C9C10=CC=CC=C10C11=CC=CC=C11C12=CC=CC=C12C13=CC=CC=C13C14=CC=CC=C14C15=CC=CC=C15C16=CC=CC=C16C17=CC=CC=C17C18=CC=CC=C18C19=CC=CC=C19C20=CC=CC=C20C21=CC=CC=C21C22=CC=CC=C22C23=CC=CC=C23C24=CC=CC=C24C25=CC=CC=C25C26=CC=CC=C26C27=CC=CC=C27C28=CC=CC=C28C29=CC=CC=C29C30=CC=CC=C30C31=CC=CC=C31C32=CC=CC=C32C33=CC=CC=C33C34=CC=CC=C34C35=CC=CC=C35C36=CC=CC=C36C37=CC=CC=C37C38=CC=CC=C38C39=CC=CC=C39C40=CC=CC=C40C41=CC=CC=C41C42=CC=CC=C42C43=CC=CC=C43C44=CC=CC=C44C45=CC=CC=C45C46=CC=CC=C46C47=CC=CC=C47C48=CC=CC=C48C49=CC=CC=C49C50=CC=CC=C50C51=CC=CC=C51C52=CC=CC=C52C53=CC=CC=C53C54=CC=CC=C54C55=CC=CC=C55C56=CC=CC=C56C57=CC=CC=C57C58=CC=CC=C58C59=CC=CC=C59C60=CC=CC=C60C61=CC=CC=C61C62=CC=CC=C62C63=CC=CC=C63C64=CC=CC=C64C65=CC=CC=C65C66=CC=CC=C66C67=CC=CC=C67C68=CC=CC=C68C69=CC=CC=C69C70=CC=CC=C70C71=CC=CC=C71C72=CC=CC=C72C73=CC=CC=C73C74=CC=CC=C74C75=CC=CC=C75C76=CC=CC=C76C77=CC=CC=C77C78=CC=CC=C78C79=CC=CC=C79C80=CC=CC=C80C81=CC=CC=C81C82=CC=CC=C82C83=CC=CC=C83C84=CC=CC=C84C85=CC=CC=C85C86=CC=CC=C86C87=CC=CC=C87C88=CC=CC=C88C89=CC=CC=C89C90=CC=CC=C90C91=CC=CC=C91C92=CC=CC=C92C93=CC=CC=C93C94=CC=CC=C94C95=CC=CC=C95C96=CC=CC=C96C97=CC=CC=C97C98=CC=CC=C98C99=CC=CC=C99C100=CC=CC=C100  
**VO-79**



C1=CC=C2C(=C1)C=CC=CC2  
**Asphaltene**

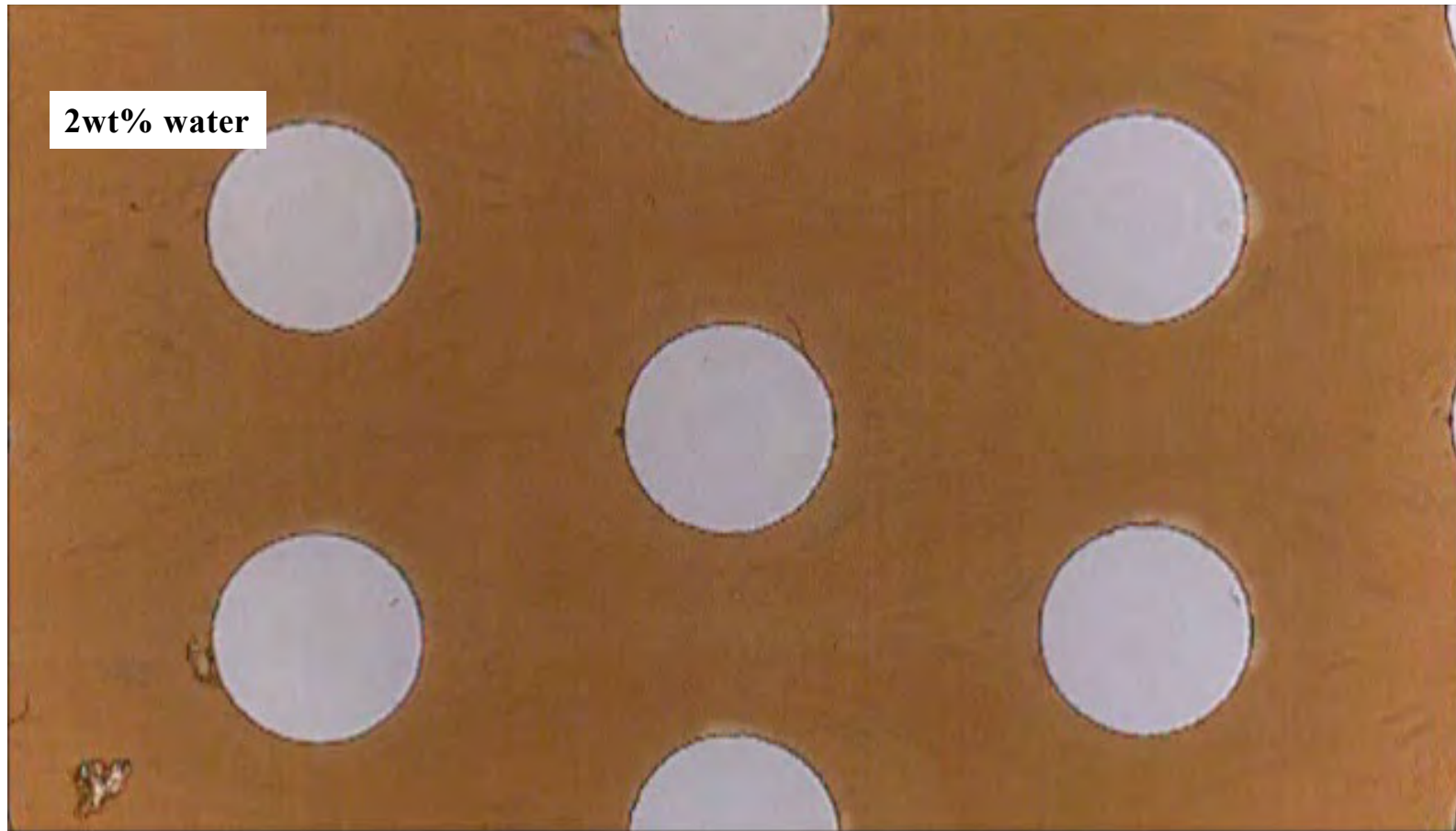


|                        | Aqueous Solution |                         |            |                         |
|------------------------|------------------|-------------------------|------------|-------------------------|
|                        | 0.5mM NaCl       | 0.5mM CaCl <sub>2</sub> | 100mM NaCl | 100mM CaCl <sub>2</sub> |
| <b>Asphaltene 1wt%</b> |                  |                         |            |                         |
| Zeta potential         | -79.50           | -45.64                  | -23.05     | 0.75                    |
| Standard deviation     | 2.16             | 2.55                    | 1.51       | 1.38                    |
| <b>VO-79 1wt%</b>      |                  |                         |            |                         |
| Zeta potential         | -60.94           | -31.10                  | -22.73     | 4.55                    |
| Standard deviation     | 4.47             | 2.50                    | 2.75       | 2.65                    |

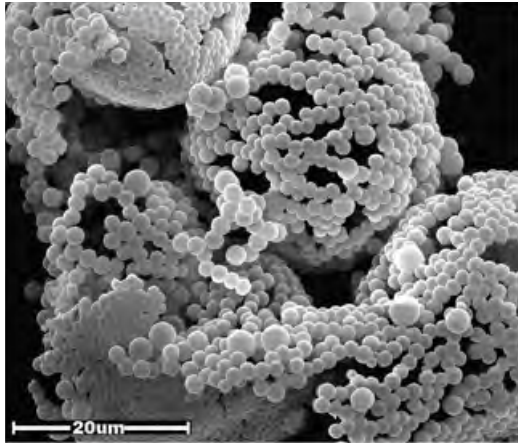
# Asphaltene Stabilized Emulsions



Flow direction

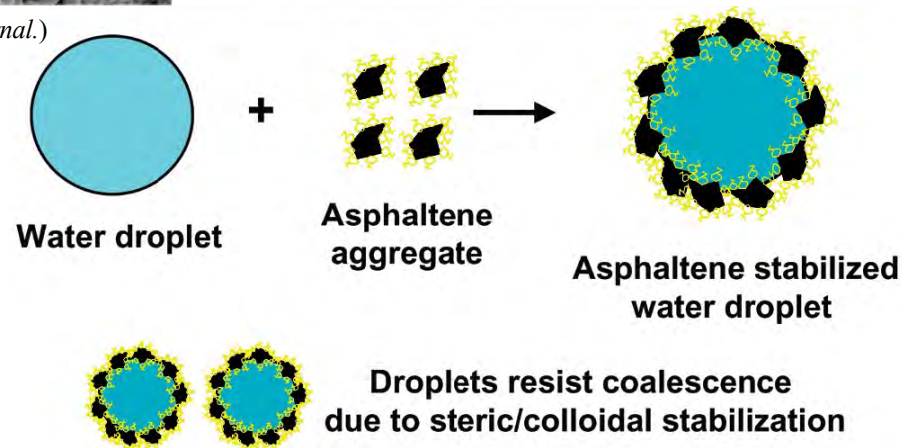


# Pickering Emulsion



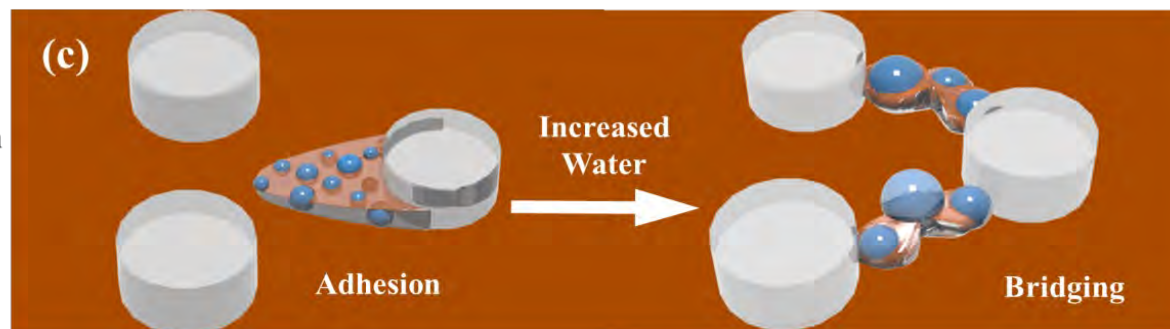
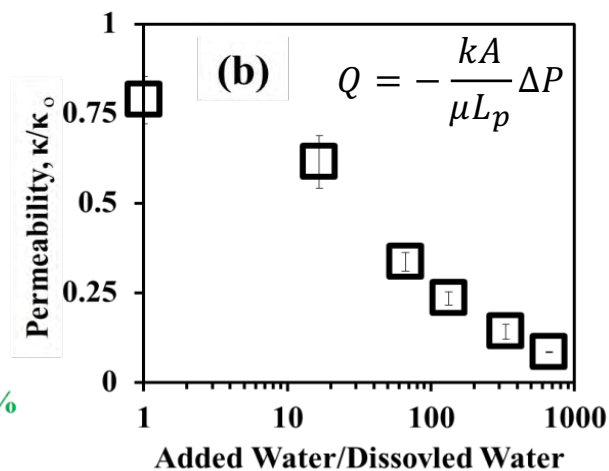
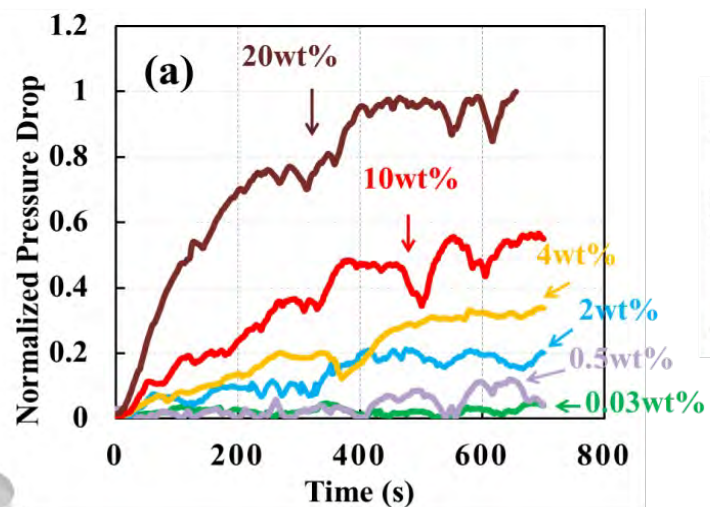
(\*Zihua Chen et al. 2013. *Polymer Journal*.)

- Particle stabilized emulsions
    - Particle laden interface is viscoelastic
    - Particles provide steric repulsion
    - Particles provide mechanical protection
- (\*Oldroyd, J.G. *Proc. Royal Soc. London A* 1955)

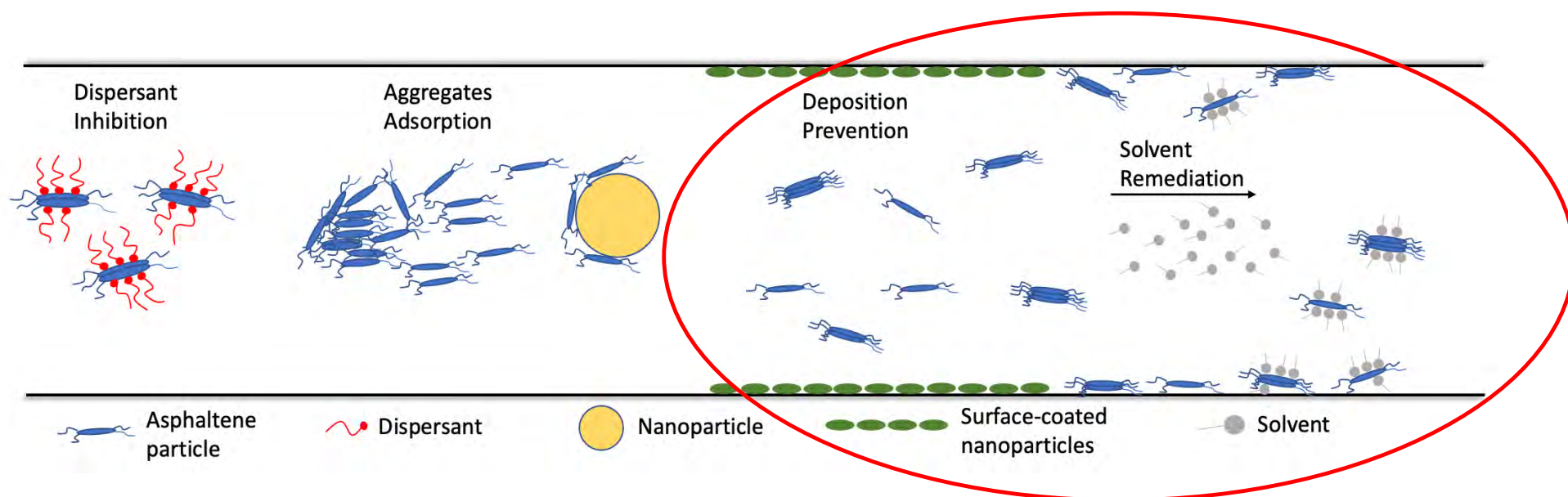


(\*Tambe, D.E et al. 1993. *J. Colloid Interface Sci.*)

# Plugging/Jamming of Asphaltene-Stabilized Emulsions



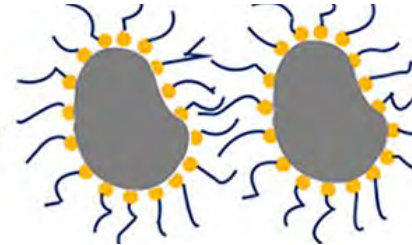
# Asphaltene Solutions



# Asphaltene Dispersants and Inhibitors



- Dispersant: suspend particles in the solution by reducing the size of the aggregates

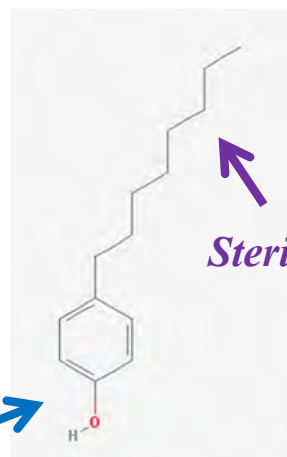


- Inhibitor: prevent asphaltene destabilization by shifting the onset of precipitation

# Asphaltene Dispersant

- Dispersant: suspend particles in the solution by reducing the size of the aggregates
- Inhibitor: prevent the destabilization and shift the onset of precipitation (usually polymer)

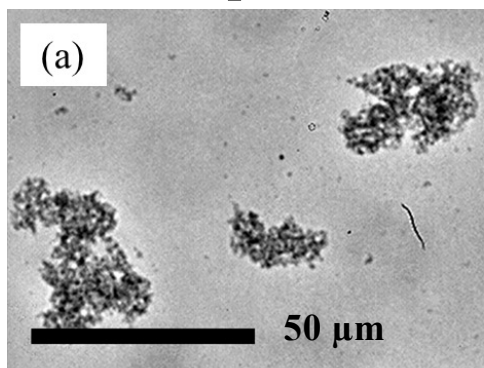
*Adsorbs on Asphaltene*



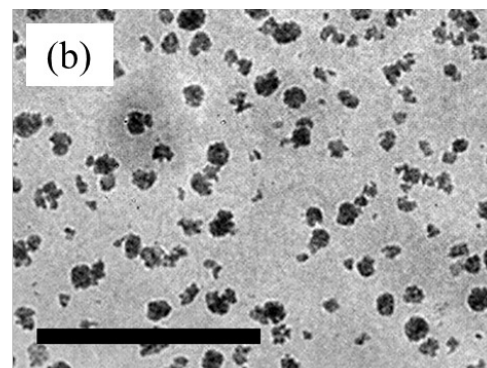
*Steric Repulsion*



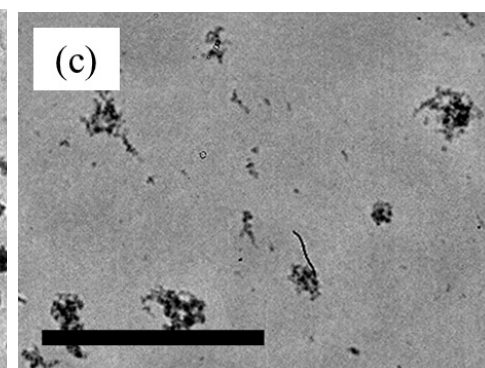
**No Dispersant**



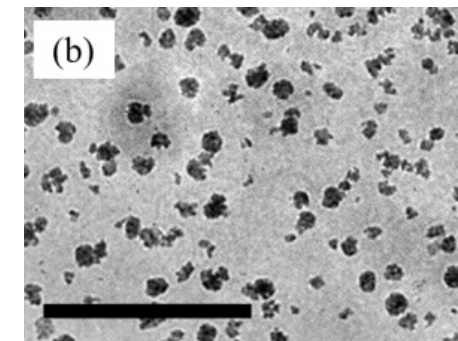
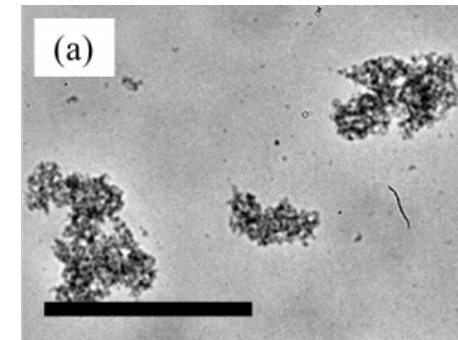
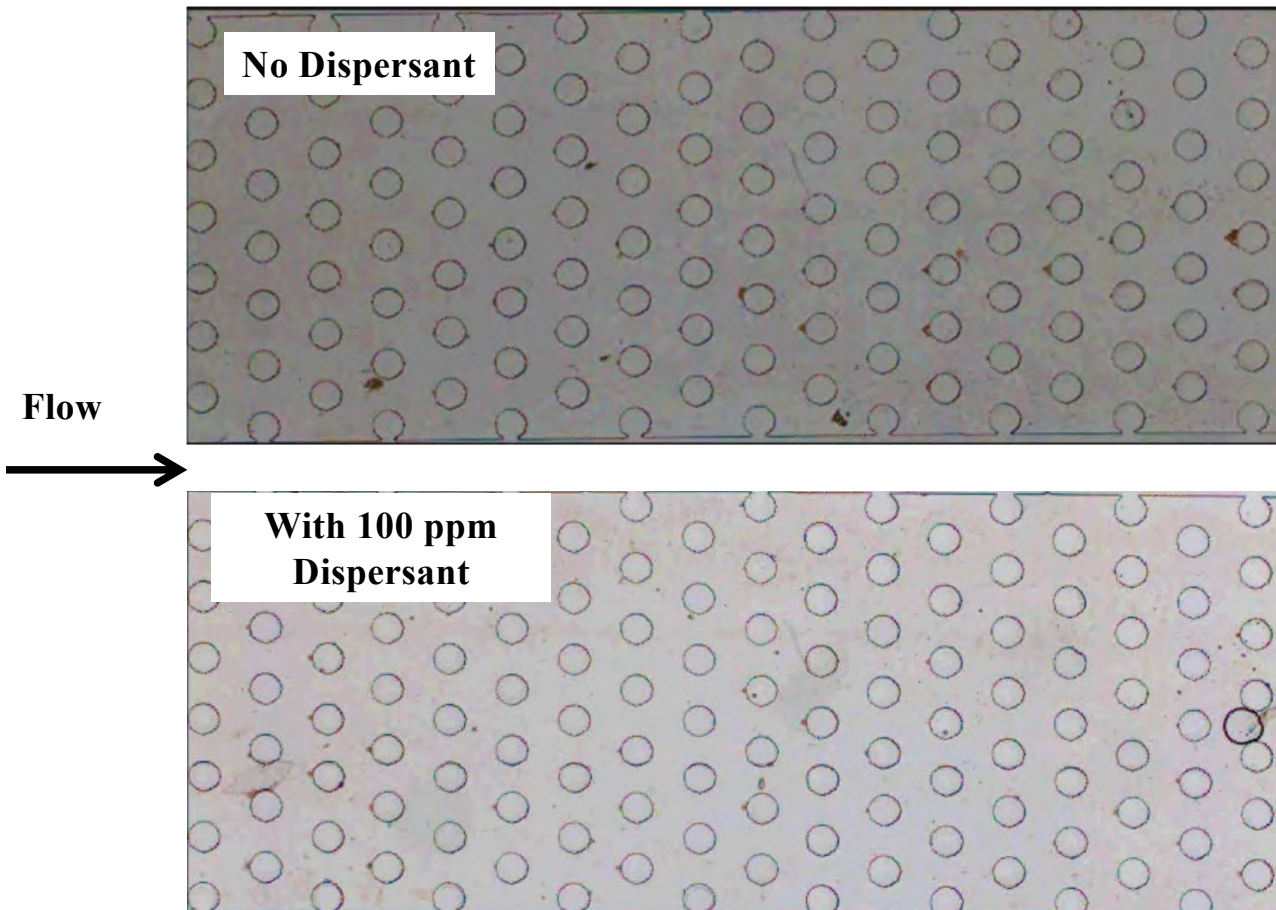
***p*-Octylphenol (pC8)**



***p*-Dodecylphenol (pC12)**

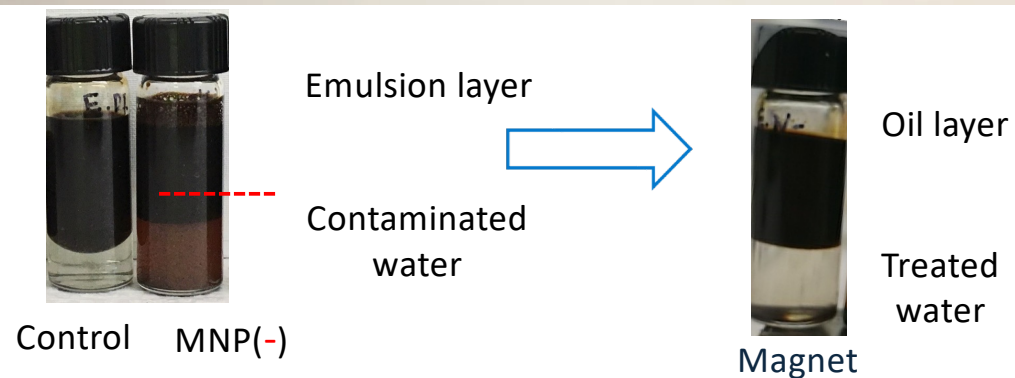
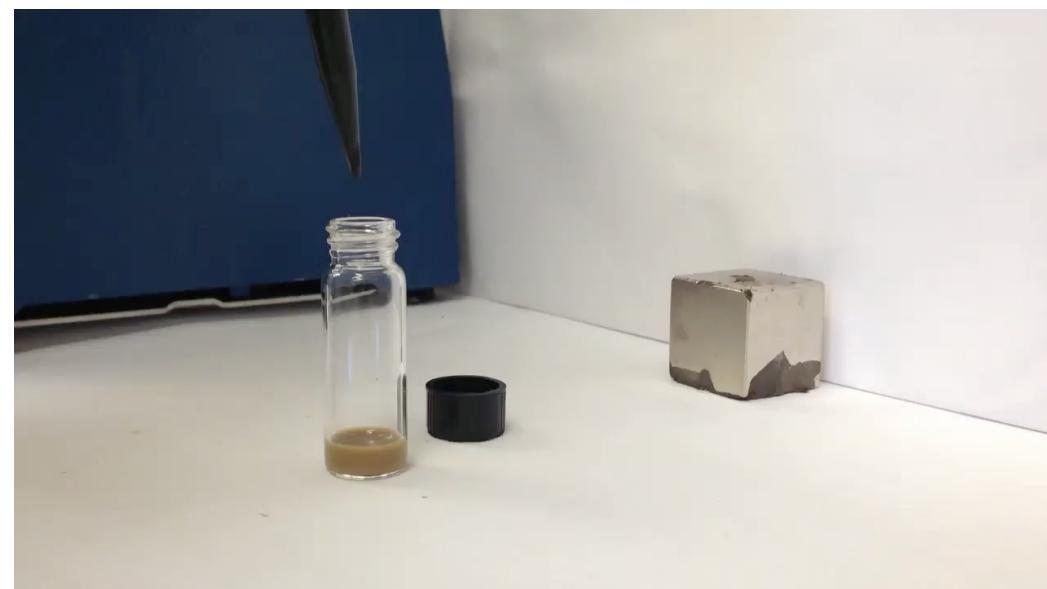
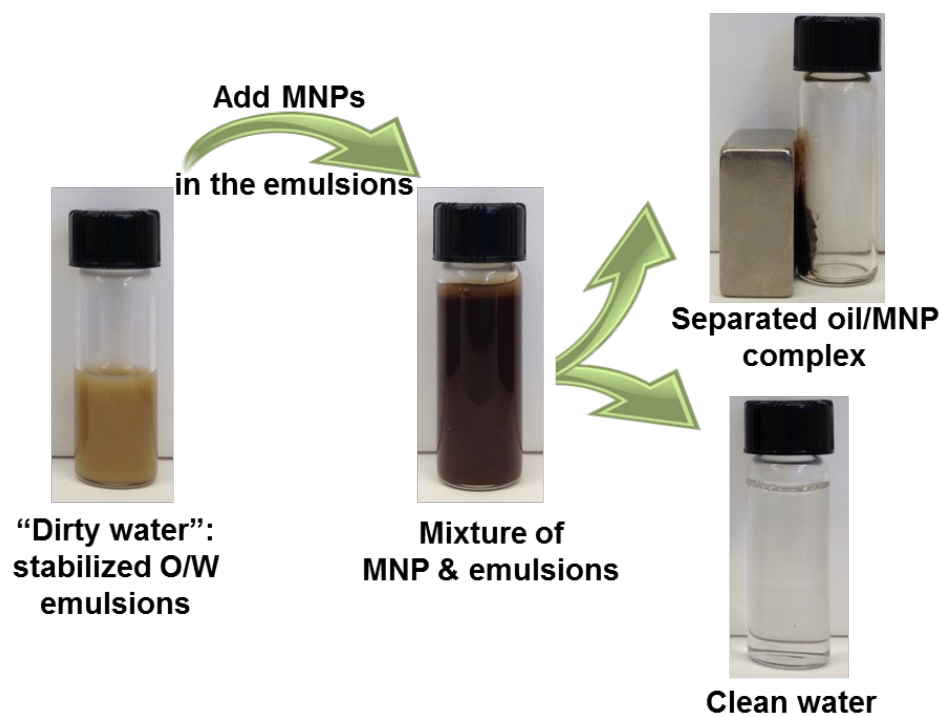


# Dynamic Deposition Process



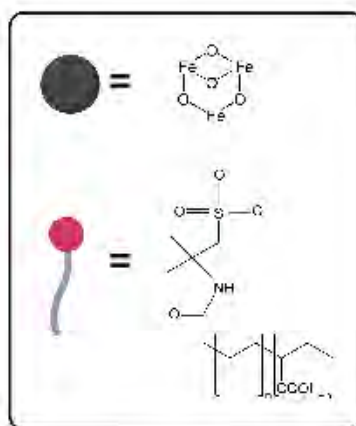
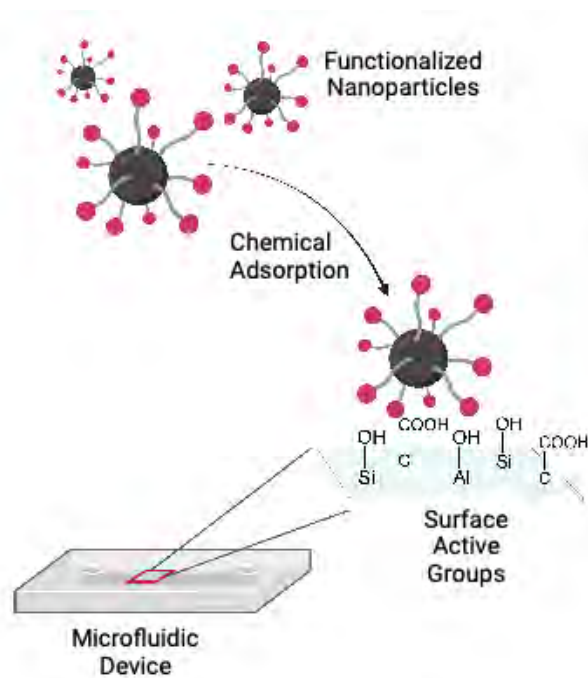
**More Deposition but “Softer” Deposits**  
➤ **Particle Interaction is Important**

# Polymer-Coated Nanoparticles

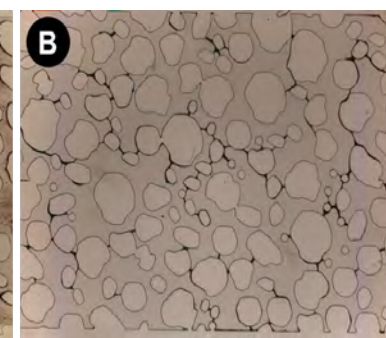
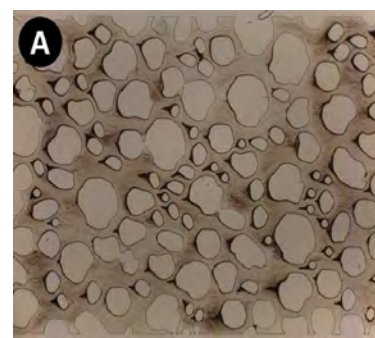


# Polymer-Coated Nanoparticles

Nanoparticles are stable in high salinity brines at reservoir conditions.

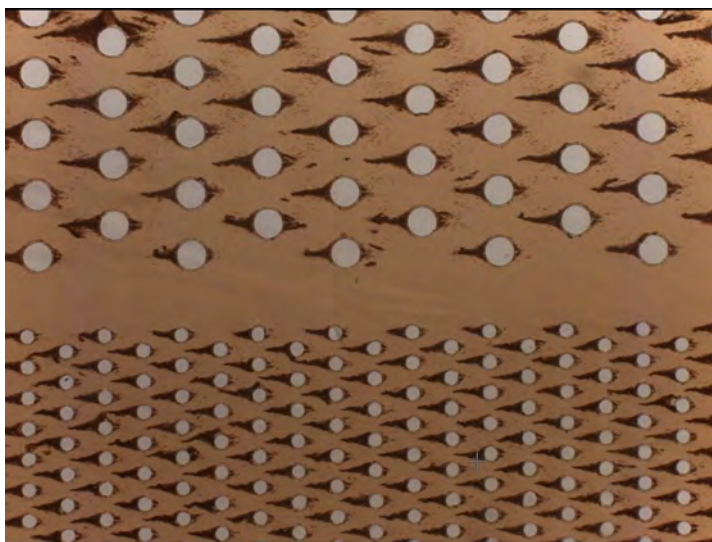


| Synthetic Brine                      |                    |
|--------------------------------------|--------------------|
| Component                            | Concentration, g/L |
| NaCl                                 | 182.31             |
| CaCl <sub>2</sub> ·2H <sub>2</sub> O | 77.25              |
| MgCl <sub>2</sub> ·6H <sub>2</sub> O | 25.62              |



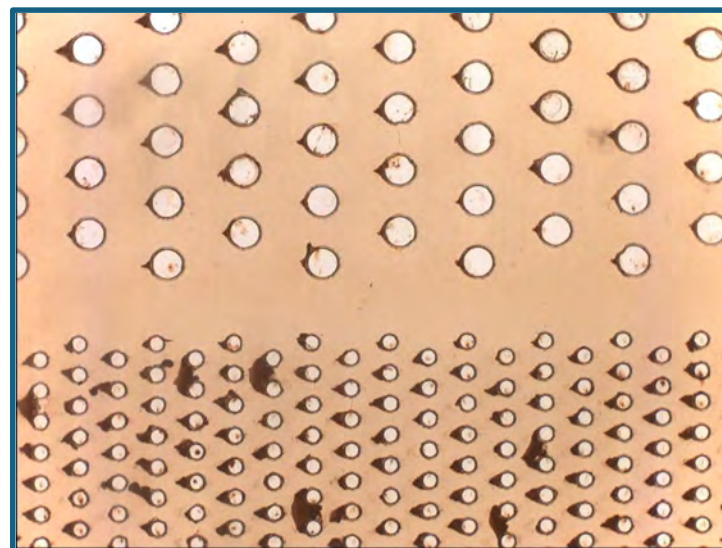
# Nanoparticle performance

Without hairy nanoparticle coating



55 Vol% Heptane - 45 Vol% Crude Oil  
@60uL/min 60 mins

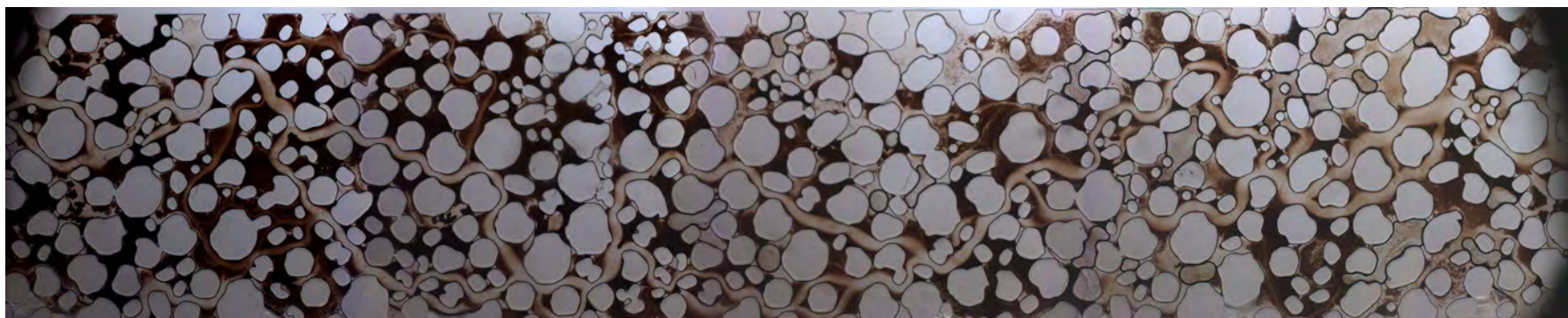
With hairy nanoparticle coating



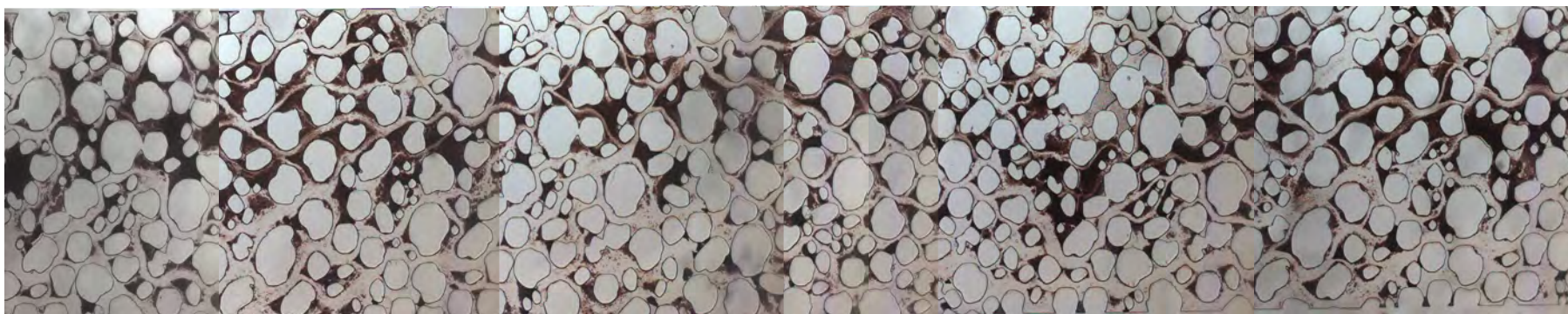
55 Vol% Heptane - 45 Vol% Crude Oil  
@60uL/min 60 mins

# Asphaltene Deposition Reduced

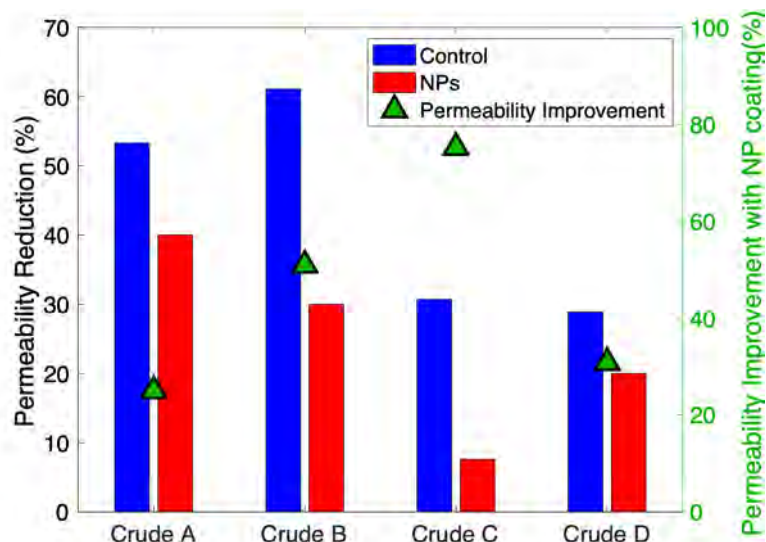
Blank Microchannel



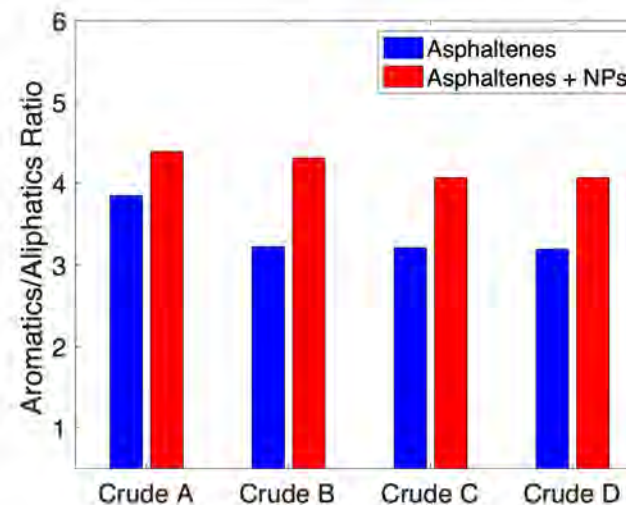
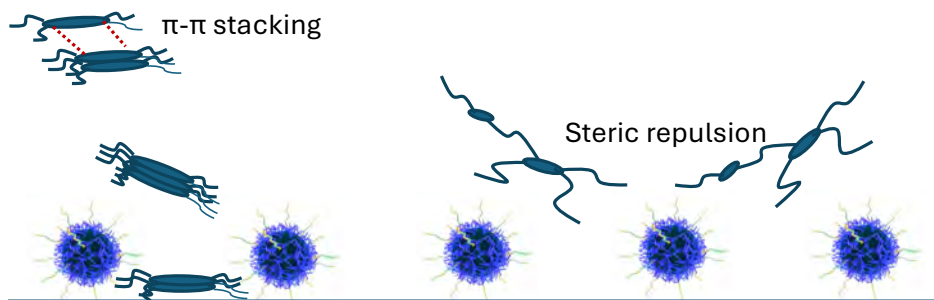
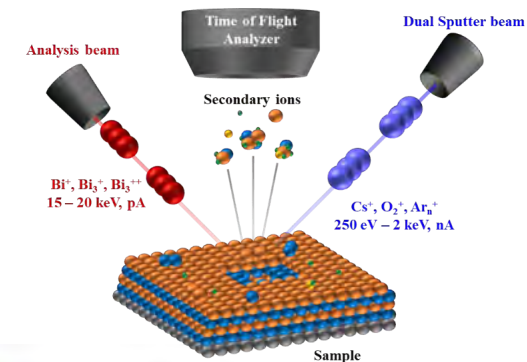
AA-AMPS-MNPs Coated Microchannel 500ppm



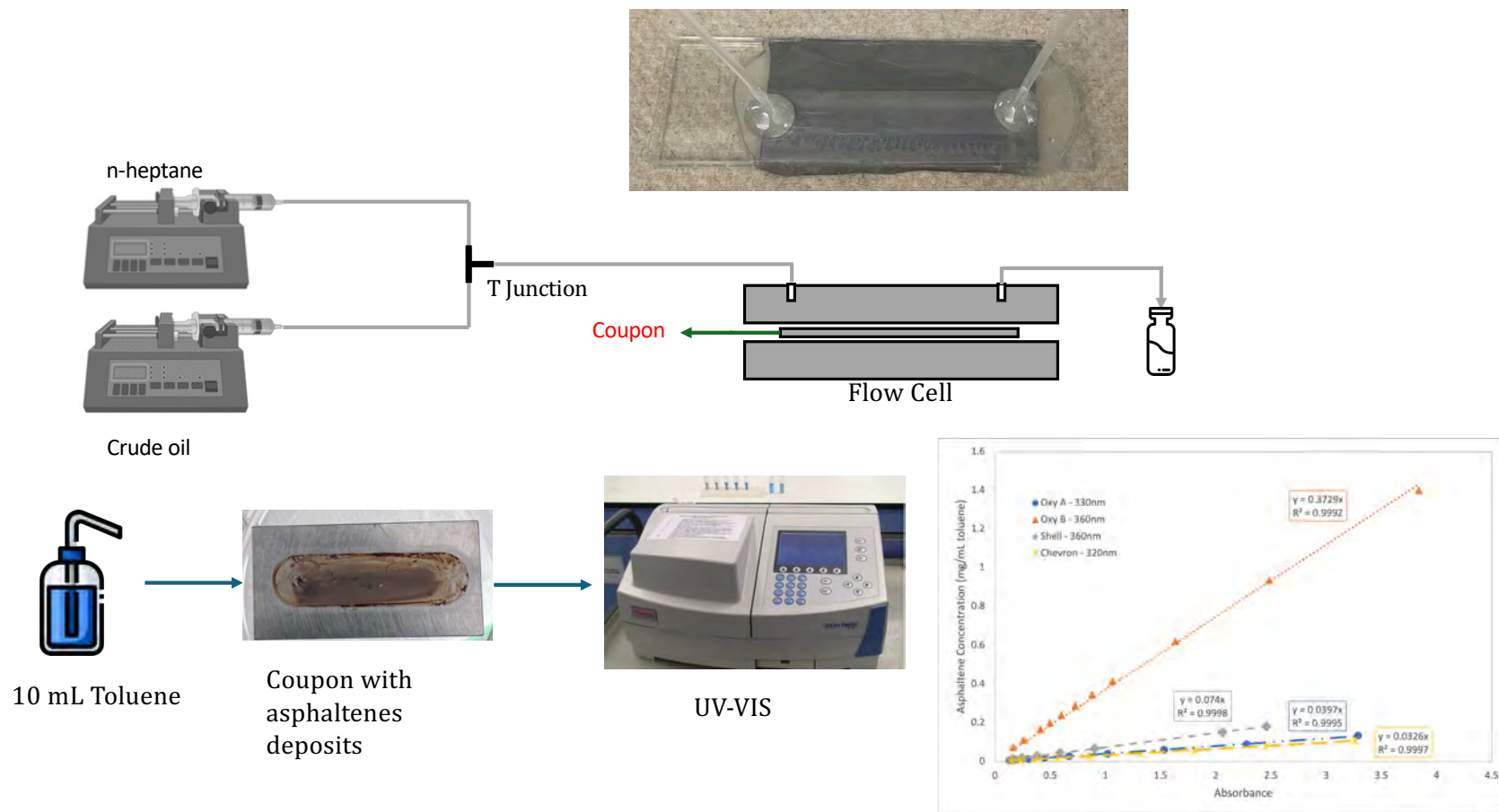
# Effectiveness of Nanoparticles by TOF-SIMS



- Time-of-Flight Secondary Ion Mass Spectrometry (ToF-SIMS) is a surface sensitive technique able to obtain both elemental composition and molecular information on the surface and in-depth.



# Asphaltene Deposition on Stainless Steel



## Images of steel

---

Control



Case # 1



Case # 2



Case # 3

MNPs coated



Case # 1

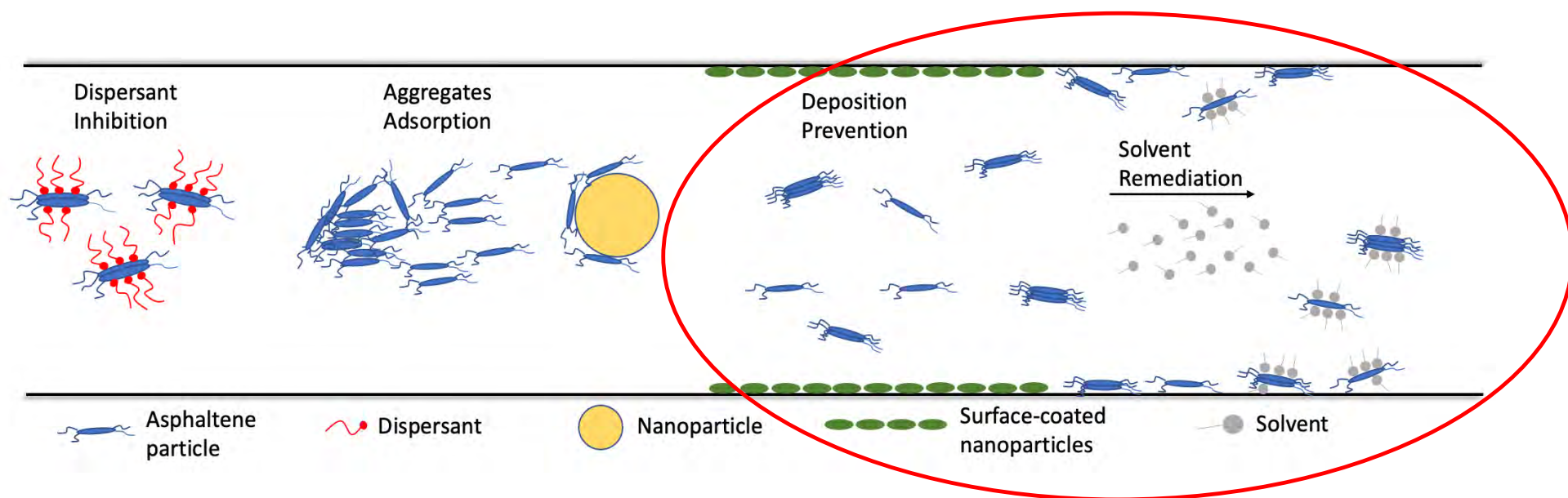


Case # 2



Case # 3

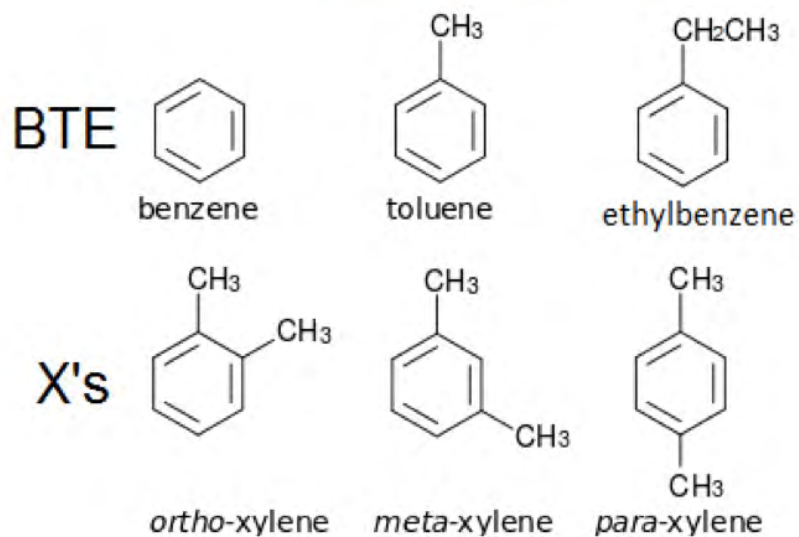
# Asphaltene Solutions



# Remediation Methods

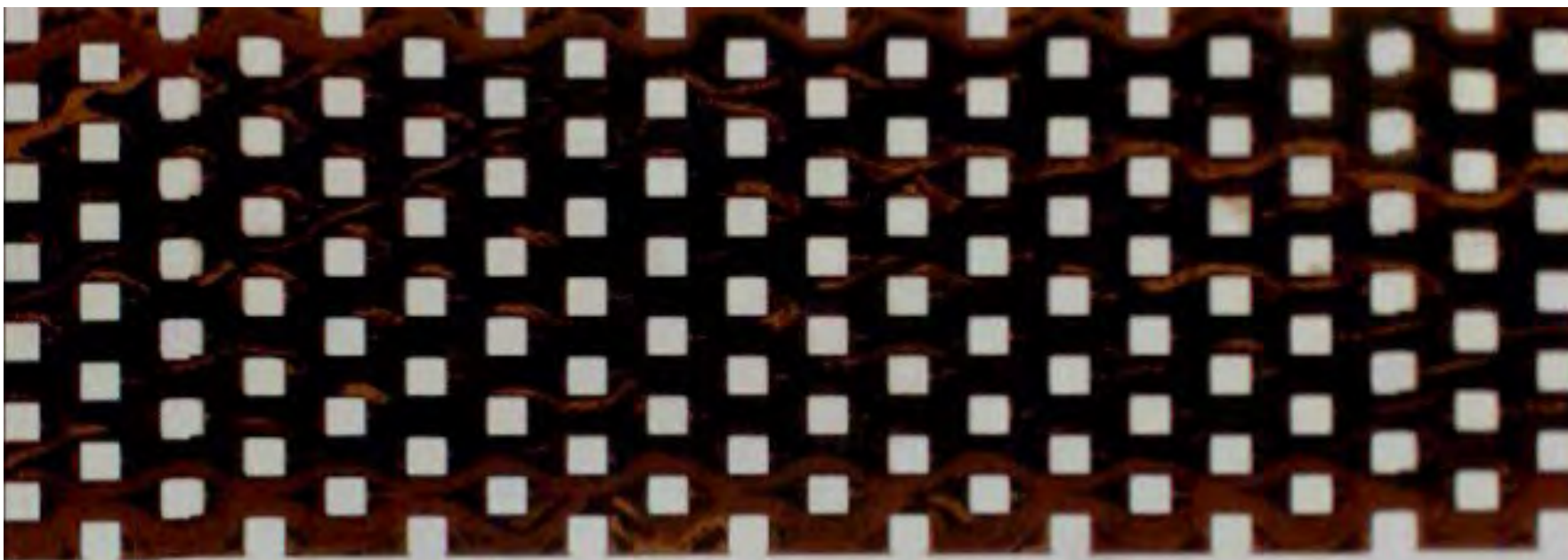


## **BTEX**



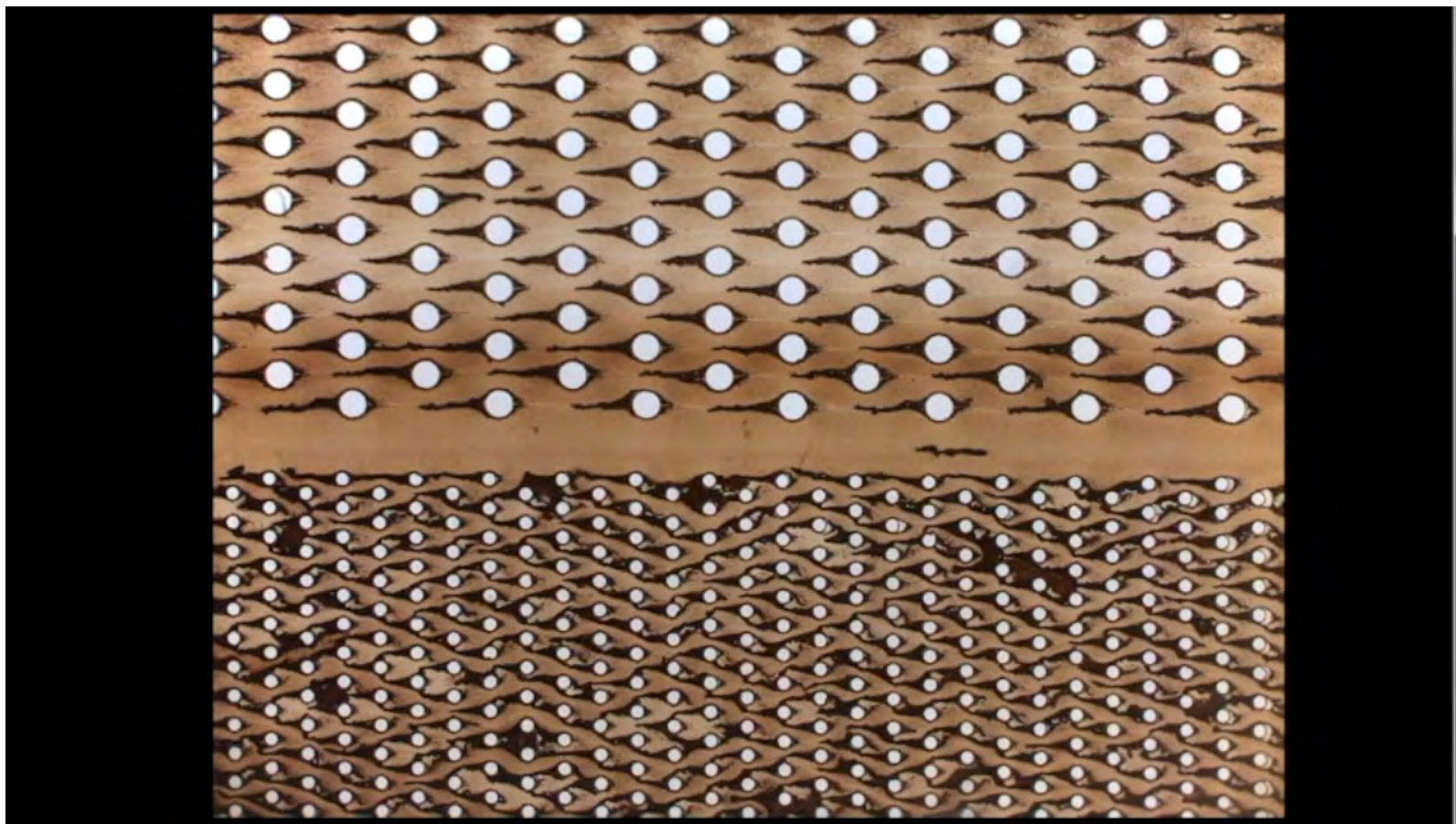
Gulf Coast Environmental  
Systems

## Washing Asphaltene Deposits with Solvents



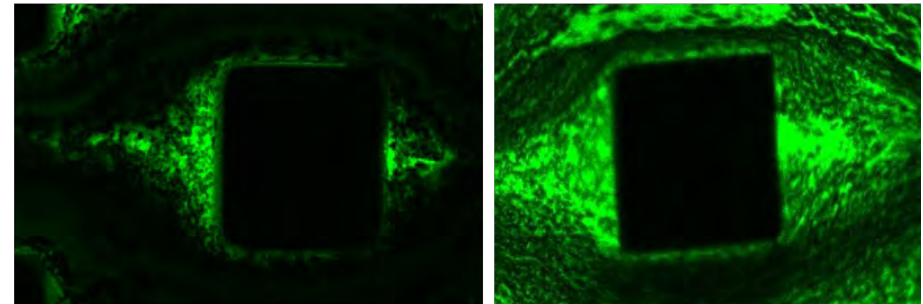
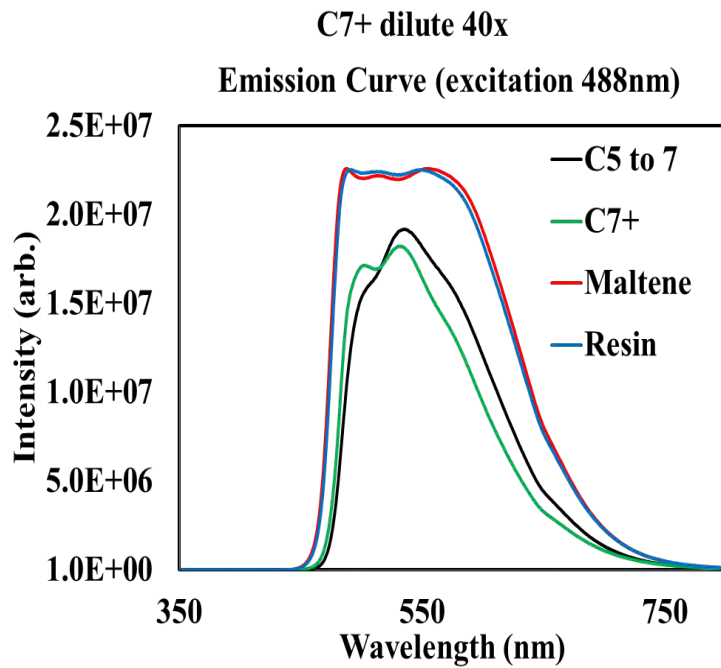
Xylene – good solvent for asphaltenes

# Washing Asphaltene Deposits with Microemulsions

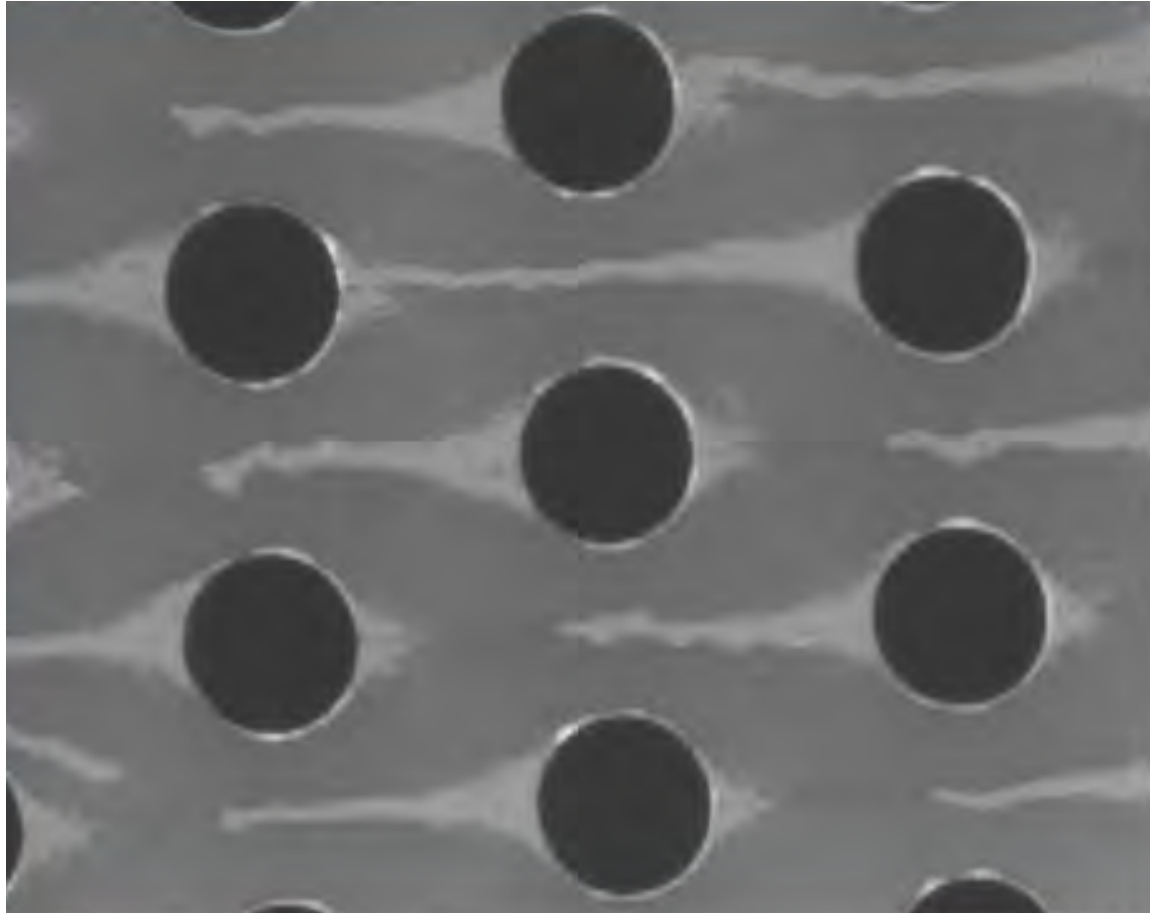


# What is being removed?

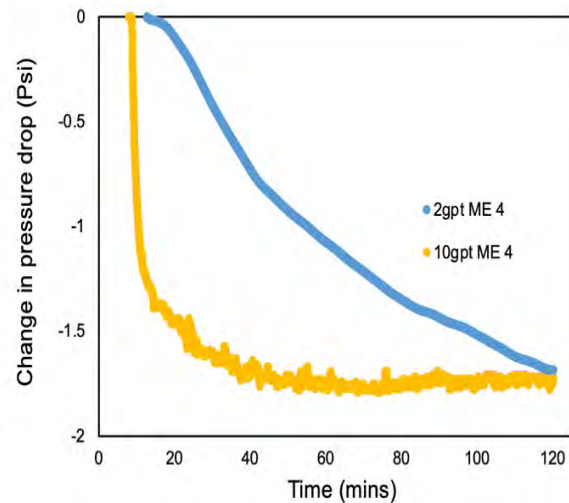
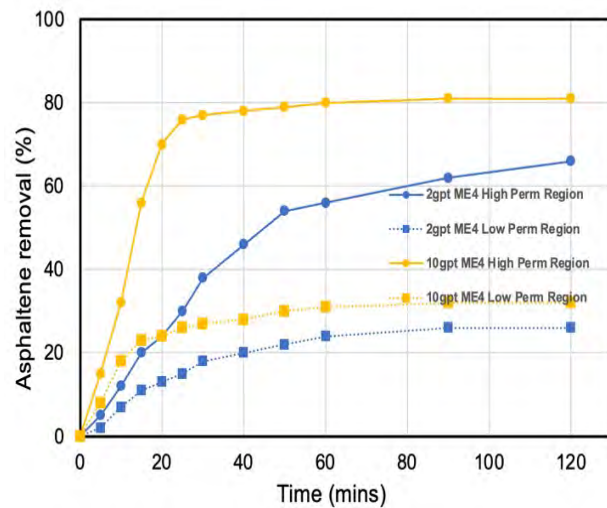
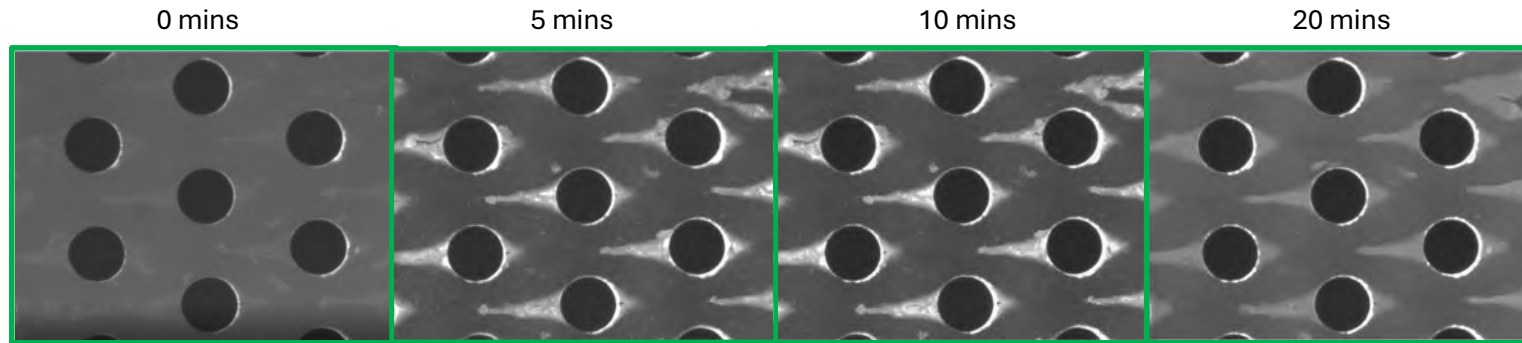
Fluorescence Microscopy Images allow Visualization of Resins (fluoresce) and Asphaltenes (non-fluorescent)



What is being removed?



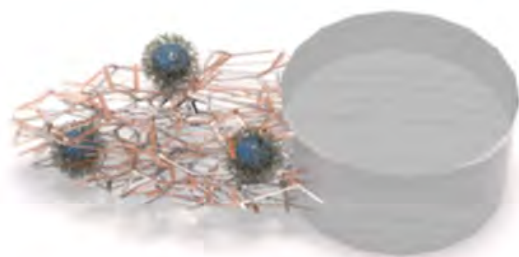
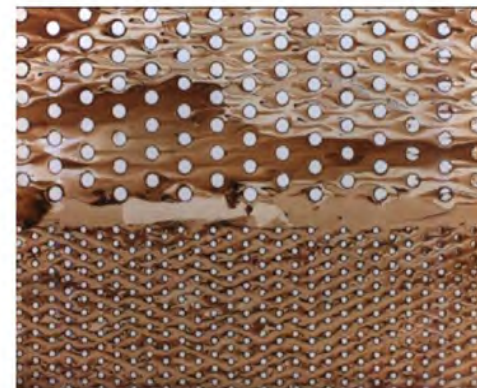
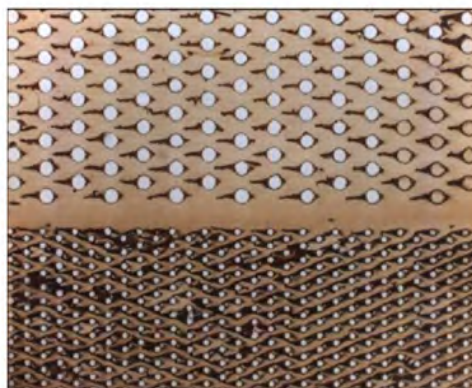
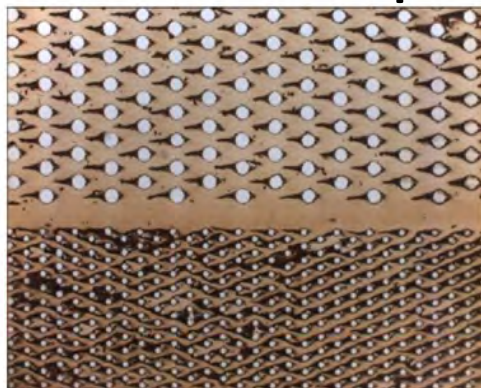
# Dynamics of asphaltene removal



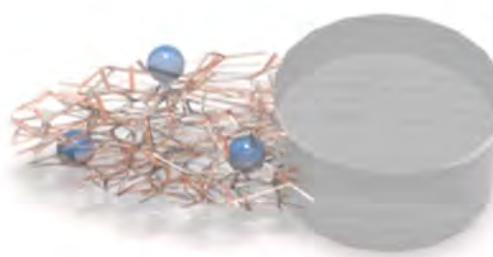
pseudo-second-order desorption kinetic model

$$\frac{t}{q_t} = \frac{1}{kq_e^2} + \frac{t}{q_e}$$

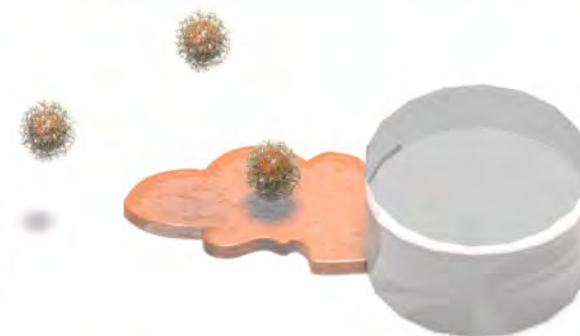
# Mechanism for asphaltene removal



Absorption of microemulsions

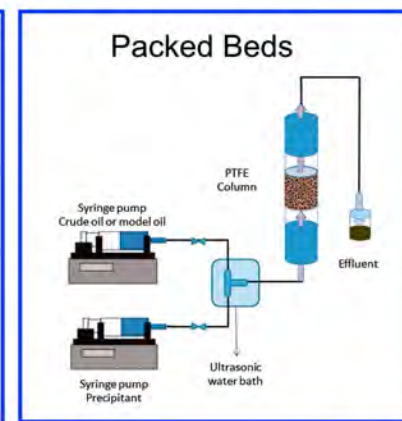
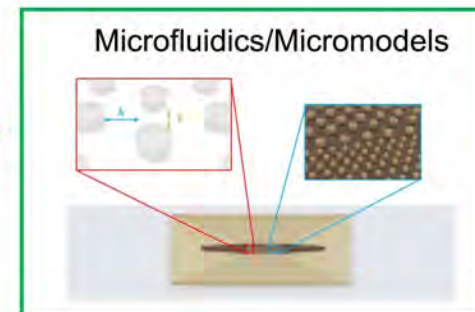
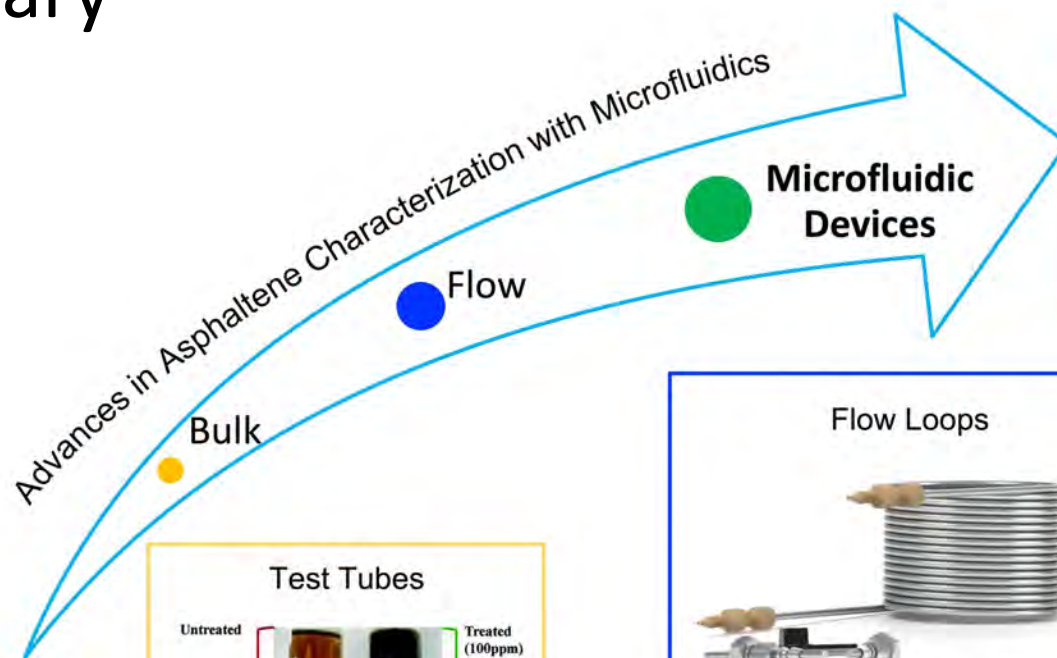


Release good solvent solubilized by microemulsions



Soften and remove asphaltene deposition

# Summary



# Acknowledgements



Langmuir (2017)  
Energy & Fuels (2017)  
JOR (2017)  
Energy & Fuels (2018)  
Energy & Fuels (2019)  
Chemical Reviews (2022)  
Energy & Fuels (2023)

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